

F332 advance notice (June 2013) *"predicted Q's"*

- How does photolysis play a part in the destruction of N_2O in the stratosphere?
- Use a series of equations to show how NO acts in the catalytic destruction of ozone?
- What is meant by the term *photolysis* ?
- Convert 100 parts per billion to a percentage, for the level of NO_x in urban areas
- Why is the $\text{N}_2\text{O}_{(\text{liquid})}$ into $\text{N}_2\text{O}_{(\text{gas})}$ **not** a true dynamic equilibrium ?
- Describe how photochemical smog can be harmful to:
(a) the environment ? (b) Humans ?
- The article mentions a reaction with HO radicals, what is a radical?
Give examples of other radicals found in the stratosphere?
- Suggest why the concentration of N_2O in the atmosphere has increased since 1820 ?
- What is used as a catalyst in the upper atmosphere in photolytic fission leading to radicals being formed ?
- Explain why N_2O is considered a greenhouse gas?
- Give the symbol equation for the reaction taking place in a catalytic converter to turn nitrogen oxides (NO_x) to oxygen and nitrogen?
Is this oxidation or reduction?
- 'Nitrous oxide is released into the atmosphere from the soil aquatic systems'. Discuss?
- Why are the three bonds in nitrogen (N_2) significant ?
- Explain using examples the mechanism of a radical chain reaction?
- What are the dangers of photochemical smog?
- Explain why lightning contributes to the formation of NO_x give one of the origins of NO_x ?
- Draw a dot and cross bonding diagram for N_2O
- Loss of N_2O by wet deposition contributes to which atmospheric problem?
- Give the oxidation state of **N** in each of the following:
a) N_2O b) HNO_3 c) NO d) $(\text{NH}_4)_2\text{SO}_4$
- Use a diagram to show the bonding in Ozone (O_3)
Describe why the bonds aren't true double bonds?
- Write balanced symbol equations to represent the reactions in the use of a three-way catalyst ?
- Describe how HNO_3 is removed from that atmosphere by wet deposition ?

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RC Schank



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Table of Contents F332 June 13 Mark Scheme

1. Understanding the eBook F332 June 13 Mark Scheme
 - The Rise of Digital Reading F332 June 13 Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying F332 June 13 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F332 June 13 Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from F332 June 13 Mark Scheme
 - Personalized Recommendations
 - F332 June 13 Mark Scheme User Reviews and Ratings
 - F332 June 13 Mark Scheme and Bestseller Lists
5. Accessing F332 June 13 Mark Scheme Free and Paid eBooks
 - F332 June 13 Mark Scheme Public Domain eBooks
 - F332 June 13 Mark Scheme eBook Subscription Services
 - F332 June 13 Mark Scheme Budget-Friendly Options
6. Navigating F332 June 13 Mark Scheme eBook Formats

- ePub, PDF, MOBI, and More
- F332 June 13 Mark Scheme Compatibility with Devices
- F332 June 13 Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of F332 June 13 Mark Scheme
 - Highlighting and Note-Taking F332 June 13 Mark Scheme
 - Interactive Elements F332 June 13 Mark Scheme
- 8. Staying Engaged with F332 June 13 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F332 June 13 Mark Scheme
- 9. Balancing eBooks and Physical Books F332 June 13 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F332 June 13 Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine F332 June 13 Mark Scheme
 - Setting Reading Goals F332 June 13 Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F332 June 13 Mark Scheme
 - Fact-Checking eBook Content of F332 June 13 Mark Scheme
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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