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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3 655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 000731 moles making the experimental molar mass acid base titration chemistry 1210 lab report containing an - Jul 15 2023

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hcl aq naoh aq nacl aq h2o l hcl a q naoh a q nacl a q h 2 o l one mole of hcl hcl would be fully neutralized by one mole of naoh naoh

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