

রাজশাহী বোর্ড

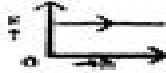
১০. পৃথিবীর কেন্দ্রে থেকে পৃথিবীর পৃষ্ঠের দিকে পড়ন্তীল বস্তুটির উপর অভিকর্ষীয় ত্বরণের সর্বোচ্চ মান কতটি?



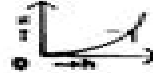
(ক)



(গ)



(খ)



১৪. কৃত্রিম উপগ্রহের পর্যায়কাল কত?
(ক) ১২ ঘণ্টা
(গ) ১ মাস

(খ) ২৪ ঘণ্টা
(ঘ) ১ বছর

১৫. পৃথিবী ১৬.৫ মিলে এবং সূর্য ৪৪ মিলে দূরত্বে একবার প্রদক্ষিণ করে, সূর্য থেকে পৃথিবীর গড় দূরত্ব $1.5 \times 10^{11} \text{ m}$ হলে সূর্য থেকে বুধের গড় দূরত্ব কত?
(ক) $128.2 \times 10^{10} \text{ m}$
(খ) $38.68 \times 10^{10} \text{ m}$
(গ) $5.81 \times 10^{10} \text{ m}$
(ঘ) $1.77 \times 10^{10} \text{ m}$

১৬. ধ্রুতি বস্তু মতো দ্রুত তিনগল হলে মহাকর্ষীয় বল হবে—



(ক) $\frac{1}{9}$ গল

(খ) $\frac{1}{3}$ গল



(গ) ১ গল

(ঘ) ৯ গল

১৭. নিচের কোন রাসিটি অ্যামোনিয়াম রাসি?
(ক) বিকৃতি
(খ) পিড্রস
(গ) ইডং ওপাক
(ঘ) পিড্রস প্রবক

১৮. তরঙ্গকালে পদার্থের অনুপাত এর মান—



(ক) -১ থেকে ০.৫



(খ) ০.৫ থেকে ১



(গ) -১ থেকে +১



(ঘ) ০.৫ থেকে ১.৫

১৯. নিচের কোনটির সর্বাধিক ক্ষতিগ্রস্ততা?



(ক) সীসা

(খ) তামা



(গ) নিকেল

(ঘ) অ্যালুমিনিয়াম

২০. একটি সরলসোলাকে কার্যকরী সৈরী তিনগল করে পৃথিবীর কেন্দ্রে নিয়ে গেলে এর দোলনকাল—



(ক) শূন্য হবে



(খ) একই থাকবে



(গ) $\sqrt{3}$ গল হবে

(ঘ) অসীম হবে

২১. একটি সরল ছন্দিত স্পন্দনে সর্বোচ্চ কালের সমীকরণ

$$y = 10 \sin \left(12\pi - \frac{\pi}{6} \right) \text{ m, কালের সর্বোচ্চ বেশ কত হবে?}$$

(ক) 1440 ms^{-1}

(খ) 120 ms^{-1}

(গ) 12 ms^{-1}

(ঘ) 10 ms^{-1}

২২. তিনটি প্যাস অকুর বেশ ক্রমক্রমে 10 ms^{-1} , 15 ms^{-1} এবং 20 ms^{-1} এর বেগ গড় করলে কত?

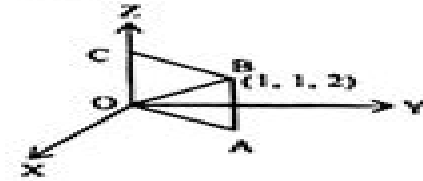
(ক) 3.87 ms^{-1}

(খ) 15 ms^{-1}

(গ) 15.54 ms^{-1}

(ঘ) 241.66 ms^{-1}

২৩.



OABC ত্রিভুজের ক্ষেত্রফল—

(ক) $\frac{1}{\sqrt{2}}$

(খ) $\sqrt{2}$

(গ) ২

(ঘ) $2\sqrt{2}$

২৪. সরলসোলায় বসে থাকা ক্ষেত্রে—

- পৃষ্ঠ থেকে যেটি কাল শূন্য নয়
- পড়ন্তের ওপর নির্ভর করে না
- কৃত্রিম উপগ্রহে আদর্শ পিড্রস হল

নিচের কোনটি সঠিক?

(ক) I ও II

(খ) I ও III

(গ) II ও III

(ঘ) I, II ও III

২৫.

কোন অবস্থায় বাতাস প্যাস আদর্শ প্যাস হিসেবে আচরণ করবে?

(ক) নিম্ন তাপ ও উচ্চ তাপমাত্রা

(খ) উচ্চ তাপ ও নিম্ন তাপমাত্রা

(গ) নিম্ন তাপ ও নিম্ন তাপমাত্রা

(ঘ) উচ্চ তাপ ও উচ্চ তাপমাত্রা

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1 calculer le potentiel de chaque électrode 2 indiquer l anode la cathode et donner les demi réactions d oxydoréduction qui
ont lieu dans chacune de ces électrodes

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équilibre suivants indiquer si l'équilibre est homogène ou hétérogène écrire l'expression de K_c écrire l'expression de K_p
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c le pH de la solution vaut 10 85 1 2 quelle est la concentration c de la

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