

Mixed Gas Laws Worksheet

- 1) How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?

$$\begin{aligned} \text{vol}^m &= 98 \text{ L} \\ P &= 2.8 \text{ atm} \\ t &= 292 \text{ K} \\ \text{mole} &= ? \\ R &= 0.082 \frac{\text{L atm}}{\text{mol K}} \end{aligned}$$

$$\begin{aligned} PV &= nRT \\ n &= \frac{PV}{RT} \\ &= \frac{2.8 \times 98}{0.082 \times 292} = \underline{\underline{11.46 \text{ mole}}} \end{aligned}$$

- 2) If 5.0 moles of O_2 and 3.0 moles of N_2 are placed in a 30.0 L tank at a temperature of 25°C , what will the pressure of the resulting mixture of gases be?

$$\begin{aligned} &5 \text{ moles } \text{O}_2 \\ &3 \text{ moles } \text{N}_2 \\ \text{vol}^m &= 30 \text{ L} \end{aligned}$$

$$\begin{aligned} \text{temp} &= 25^\circ\text{C} \rightarrow 273 + 25 \\ &= \underline{\underline{298 \text{ K}}} \\ P &= ? \end{aligned}$$

$$\begin{aligned} PV &= nRT \\ P &= \frac{nRT}{V} = \frac{8 \times 0.082 \times 298}{30 \text{ L}} \\ &= \underline{\underline{6.5 \text{ atm}}} \end{aligned}$$

- 3) A balloon is filled with 35.0 L of helium in the morning when the temperature is 20.0°C . By noon the temperature has risen to 45.0°C . What is the new volume of the balloon?

$$\begin{aligned} \textcircled{1} \text{ Vol}^m &= 35 \text{ L} \\ \textcircled{1} \text{ Temp} &= 20^\circ\text{C} = 293 \text{ K} \\ \textcircled{2} \text{ Temp} &= 45^\circ\text{C} = 318 \text{ K} \\ \textcircled{2} \text{ Vol}^m &= ? \end{aligned}$$

$$\begin{aligned} &\text{Charles law} \\ \frac{V_1}{T_1} &= \frac{V_2}{T_2} \\ \frac{35}{293} &= \frac{V_2}{318} \\ V_2 &= \frac{35 \times 318}{293} = \underline{\underline{37.78 \text{ L}}} \end{aligned}$$

- 4) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

$$\begin{aligned} V &= 35 \text{ L} \\ T &= 315 \text{ K} \\ P &= 190 \text{ atm} \\ n &= ? \end{aligned}$$

$$\begin{aligned} PV &= nRT \\ n &= \frac{PV}{RT} = \frac{190 \times 35}{0.082 \times 315} \\ &= \underline{\underline{257.45 \text{ mol}}} \end{aligned}$$

Mixed Gas Laws Key

S Nieto

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web in this paper several methods for identification characterization tests of organic compounds determination of molecular weight solubility of unknown compounds tlc technique each of you will receive two vials one containing a liquid unknown and one a solid unknown

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web this lab report is based upon unknown binary mixture e containing unknowns 6s and 15l as it contains a solid liquid mixture vacuum filtration was conducted 6s is insoluble in 15l procedure 1 take the unknown binary mixture and make observations about the mixture by stating its state odor color and solubility characteristics