

The Hydrogen Atom – Student Challenge

Background Material

Classically, read the background information on Energy Levels, Light, and Transitions, and answer the following questions on these (your understanding)

Operation 1: I completed the following table which compares how the Bohr Model and the Quantum Mechanical approach the Hydrogen atom. In what "interesting" facts describe things in the model you think are interesting that it does.

Concept	Bohr Model	Quantum Model
Electrons	positively charged particles in the center	positively charged proton in the center
Electrons	orbiting electrons	negatively charged cloud
Orbitals	Electrons do not move in any the orbitally paths	Electrons moving around central nuclei
Transitions of electrons in the atom orbital	moving a electron to the Bohr orbital	Electron wave
Energy of the electron levels	$E_1 = -13.6 \text{ eV}$ $E_2 = -3.4 \text{ eV}$	It is different

Operation 2: I completed the following table which compares the properties of two different elements by adding their appropriate relationships. What does this is completed for you: the red photon has a longer wavelength, smaller frequency, smaller energy, and the same velocity. How do you see in adding photon?

Hydrogen Atom Student Guide Solutions Naap

L Cohen



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Embracing the Track of Appearance: An Psychological Symphony within **Hydrogen Atom Student Guide Solutions Naap**

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