

## Three Hydrogen-like Atoms – Simultaneous Quantities

### Electromagnetic Radiation

Consider now the hydrogen atom, which emits energy levels, light, and transitions and also is the following quantities in terms of its understanding:

Quantum is a concept that describes the behavior of particles, such as the photon, which is the quantum of electromagnetic radiation. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom.

Quantity	Hydrogen Atom	Quantum Number
Energy	Discrete energy levels, given by the formula $E_n = -13.6 \text{ eV} / n^2$ , where $n$ is the principal quantum number.	Discrete energy levels, given by the formula $E_n = -13.6 \text{ eV} / n^2$ , where $n$ is the principal quantum number.
Angular momentum	Discrete angular momentum values, given by the formula $L = \hbar \sqrt{l(l+1)}$ , where $l$ is the orbital angular momentum quantum number.	Discrete angular momentum values, given by the formula $L = \hbar \sqrt{l(l+1)}$ , where $l$ is the orbital angular momentum quantum number.
Transition frequency	Discrete transition frequencies, given by the formula $\nu = (E_i - E_f) / h$ , where $E_i$ and $E_f$ are the initial and final energy levels, and $h$ is Planck's constant.	Discrete transition frequencies, given by the formula $\nu = (E_i - E_f) / h$ , where $E_i$ and $E_f$ are the initial and final energy levels, and $h$ is Planck's constant.
Energy of the photon	Discrete energy values, given by the formula $E = h\nu$ , where $\nu$ is the frequency of the photon.	Discrete energy values, given by the formula $E = h\nu$ , where $\nu$ is the frequency of the photon.

Quantum is a concept that describes the behavior of particles, such as the photon, which is the quantum of electromagnetic radiation. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom.

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**Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

**Student Solutions Manual for General Chemistry** John McMurry, Robert Fay, 2013-03-07 By Joseph Topich Virginia Commonwealth University This manual for students contains solutions to selected all in chapter problems and even numbered end of chapter problems

**Study Guide and Solutions Manual to Accompany Organic Chemistry, 11th Edition** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2013-03-25 This is the study guide and solutions manual to accompany Organic Chemistry 11th Edition

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