

Highlight ALL Answers

Student Exploration: Covalent Bonds



Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? They can each share one of there markers with eachother, so they have a full set.

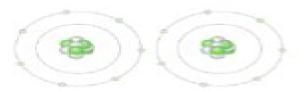
Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They should share the colors that they are missing

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play () to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have? 7 valence electrons
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play.
 What happens? A shell is added, and it connects 2 atoms.
- Click Pause, drag an electron from the right atom to the left, and then click Play.What happens now? An electron is added to the shell connecting the atoms.



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IM Harris

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Table of Contents Gizmo Student Exploration Sheet Answers Covalent Bonds

- 1. Understanding the eBook Gizmo Student Exploration Sheet Answers Covalent Bonds
 - The Rise of Digital Reading Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - $\circ \ \ Determining \ Your \ Reading \ Goals$
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Student Exploration Sheet Answers Covalent Bonds
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Personalized Recommendations

- Gizmo Student Exploration Sheet Answers Covalent Bonds User Reviews and Ratings
- Gizmo Student Exploration Sheet Answers Covalent Bonds and Bestseller Lists
- 5. Accessing Gizmo Student Exploration Sheet Answers Covalent Bonds Free and Paid eBooks
 - Gizmo Student Exploration Sheet Answers Covalent Bonds Public Domain eBooks
 - Gizmo Student Exploration Sheet Answers Covalent Bonds eBook Subscription Services
 - Gizmo Student Exploration Sheet Answers Covalent Bonds Budget-Friendly Options
- 6. Navigating Gizmo Student Exploration Sheet Answers Covalent Bonds eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Gizmo Student Exploration Sheet Answers Covalent Bonds Compatibility with Devices
 - Gizmo Student Exploration Sheet Answers Covalent Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Highlighting and Note-Taking Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Interactive Elements Gizmo Student Exploration Sheet Answers Covalent Bonds
- 8. Staying Engaged with Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploration Sheet Answers Covalent Bonds
- 9. Balancing eBooks and Physical Books Gizmo Student Exploration Sheet Answers Covalent Bonds
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploration Sheet Answers Covalent Bonds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Setting Reading Goals Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Student Exploration Sheet Answers Covalent Bonds
 - Fact-Checking eBook Content of Gizmo Student Exploration Sheet Answers Covalent Bonds

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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