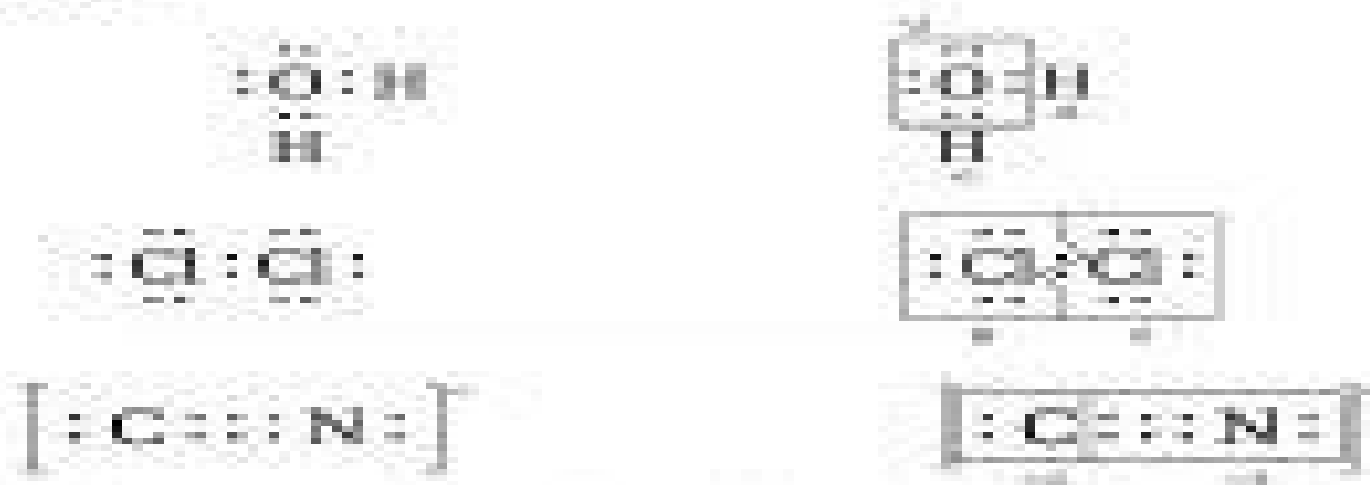


Resonance-Structural Drawing

Resonance Structures:



1. correct electron-distribution, the number of valence electrons, and the overall charge of the molecule, correct formal charges, and

2. correct Lewis arrangement of the atoms

3. correct formal charges

There should be a potential for resonance structures.

1. A molecule with resonance structures is any an electron delocalized system, or conjugated. A delocalized system is a molecule that has



2. or

There is one double bond adjacent to

another double bond.

This should be potential for resonance structures.



...resonance structures have equivalent electron density at each carbon atom and are separated by single bonds.

3. or

There are 4 conjugated double bonds or more.

4. Two adjacent double bonds

5. Conjugated or planar system conjugated system

1. Conjugated system is a system where the p-orbitals are in a continuous overlap.

2. Conjugated system is a system where the p-orbitals are in a continuous overlap.



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RJ Shavelson



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