

# Exercise 1: Kinetics

1.1. A reaction with reactants and products. Consider the reaction with the following forward and reverse rate constants:

1. No elementary reaction is faster than the reverse reaction.

2. The  $k_{\text{fwd}}$  is always  $k_{\text{rev}}$  for a given reaction.

3. The equilibrium constant is  $k_{\text{fwd}}/k_{\text{rev}}$  for a given reaction.

1.2. The reaction is reversible. Consider the reaction with the forward and reverse rate constants given below.

1. The  $k_{\text{fwd}}$  is  $k_{\text{rev}}$  for a given reaction.

2. The  $k_{\text{fwd}}$  is not a rate constant for a given reaction.

3. The  $k_{\text{rev}}$  is  $k_{\text{fwd}}$  for a given reaction.

1.3. The reaction is reversible. Consider the reaction with the forward and reverse rate constants given below. The reaction is reversible.

1. The reaction is reversible.

2. The reaction is reversible.

3. The reaction is reversible.

1.4. The reaction is reversible.

1. The reaction is reversible.

2. The reaction is reversible.

1.5. The reaction is reversible. Consider the reaction with the forward and reverse rate constants given below.

1. The reaction is reversible.

2. The reaction is reversible.

3. The reaction is reversible.

1.6. The reaction is reversible.

1.7. The reaction is reversible. Consider the reaction with the forward and reverse rate constants given below.

1. The reaction is reversible.

# Lab Manual Answers Spanish Imagina Leccion 6

**ML Yell**



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