

## Gravimetric Factor - Example

**Question:** Calculate the grams of analyte per gram of precipitate for the following conversion:

<i>Analyte</i>	<i>Precipitate</i>
$\text{Bi}_2\text{S}_3$	$\text{BaSO}_4$

**Answer:**

$$\text{g Bi}_2\text{S}_3/\text{g BaSO}_4 = \frac{\text{FW Bi}_2\text{S}_3(\text{g/mol})}{\text{FW BaSO}_4(\text{g/mol})} \times \frac{1}{3} (\text{mol Bi}_2\text{S}_3 / \text{mol BaSO}_4)$$

$$\text{GF} = \frac{514.15(\text{g Bi}_2\text{S}_3/\text{mol})}{233.40(\text{g BaSO}_4/\text{mol})} \times \frac{1}{3} = 0.73429 \text{ g Bi}_2\text{S}_3/\text{g BaSO}_4$$

# Gravimetric Analysis Lab Calculations

**JL Elias**



## **Gravimetric Analysis Lab Calculations:**

## Unveiling the Magic of Words: A Report on "**Gravimetric Analysis Lab Calculations**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Gravimetric Analysis Lab Calculations**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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