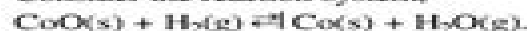


16 • Chemical Equilibrium

1. Consider the reaction system,



The equilibrium constant expression is

- a) $\frac{[\text{CoO}][\text{H}_2]}{[\text{Co}][\text{H}_2\text{O}]}$ d) $\frac{[\text{H}_2]}{[\text{H}_2\text{O}]}$
 b) $\frac{[\text{Co}][\text{H}_2\text{O}]}{[\text{CoO}][\text{H}_2]}$ c) $\frac{[\text{H}_2\text{O}]}{[\text{H}_2]}$
 e) $\frac{[\text{Co}][\text{H}_2\text{O}]}{[\text{H}_2]}$

2. Given the equilibrium,



if this equilibrium is established by beginning with equal number of moles of SO_2 and O_2 in a 1.0 Liter bulb, then the following *must* be true at equilibrium:

- a) $[\text{SO}_2] = [\text{SO}_3]$ d) $[\text{SO}_2] < [\text{O}_2]$
 b) $2[\text{SO}_2] = 2[\text{SO}_3]$ c) $[\text{SO}_2] > [\text{O}_2]$
 e) $[\text{SO}_2] = [\text{O}_2]$

Questions 3 & 4 refer to the following:

At a given temperature, 0.300 mole NO , 0.200 mol Cl_2 and 0.500 mol ClNO were placed in a 25.0 Liter container. The following equilibrium is established: $2\text{ClNO(g)} \rightleftharpoons 2\text{NO(g)} + \text{Cl}_2\text{(g)}$

3. At equilibrium, 0.600 mol of
- ClNO
- was present. The number of
- moles*
- of
- Cl_2
- present at equilibrium is

- a) 0.050 d) 0.200
 b) 0.100 e) 0.250
 c) 0.150

PRACTICE TEST

4. The equilibrium constant,
- K_c
- , is:

- a) 4.45×10^{-4} d) 0.167
 b) 6.67×10^{-4} e) 1500
 c) 0.111

5. At
- 985°C
- , the equilibrium constant for the reaction,



is 1.63. What is the equilibrium constant for the reverse reaction?

- a) 1.63 d) 0.613
 b) 0.815 e) 1.00
 c) 2.66

6. What is the relationship between
- K_p
- and
- K_c
- for the reaction,
- $2\text{ICl(g)} \rightleftharpoons \text{I}_2\text{(g)} + \text{Cl}_2\text{(g)}$
- ?

- a) $K_p = K_c(RT)^{-1}$ d) $K_p = K_c$
 b) $K_p = K_c(RT)$ e) $K_p = K_c(2RT)$
 c) $K_p = K_c(RT)^2$

7. For the reaction
- $2\text{NO}_2\text{(g)} \rightleftharpoons \text{N}_2\text{O}_4\text{(g)}$
- ,
- K_p
- at
- 25°C
- is 7.3, when all partial pressures are expressed in atmospheres. What is
- K_c
- for this reaction? [
- $R=0.0821 \text{ L}\cdot\text{atm}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$
-]

- a) 4270 d) 179
 b) 0.0119 e) 2.06
 c) 0.291

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Chemical Equilibrium William Guenther, 1975-12 The present work is designed to provide a practical introduction to aqueous equilibrium phenomena for both students and research workers in chemistry biochemistry geochemistry and interdisciplinary environmental fields The pedagogical strategy I have adopted makes heavy use of detailed examples of problem solving from real cases arising both in laboratory research and in the study of systems occurring in nature The procedure starts with mathematically complete equations that will provide valid solutions of equilibrium problems instead of the traditional approach through approximate concentrations and idealized infinite dilution assumptions There is repeated emphasis on the use of corrected conditional equilibrium constants and on the checking of numerical results by substitution in complete equations and or against graphs of species distributions Graphical methods of calculation and display are used extensively because of their value in clarifying equilibria and in leading one quickly to valid numerical approximations The coverage of solution equilibrium phenomena is not however exhaustively comprehensive Rather I have chosen to offer fundamental and rigorous examinations of homogeneous step equilibria and their interactions with solubility and redox equilibria Many examples are worked out in detail to demonstrate the use of equilibrium calculations and diagrams in various fields of investigation

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Table of Contents Equilibrium Worksheet Pasadena

1. Understanding the eBook Equilibrium Worksheet Pasadena
 - The Rise of Digital Reading Equilibrium Worksheet Pasadena
 - Advantages of eBooks Over Traditional Books
2. Identifying Equilibrium Worksheet Pasadena
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Equilibrium Worksheet Pasadena
 - User-Friendly Interface
4. Exploring eBook Recommendations from Equilibrium Worksheet Pasadena
 - Personalized Recommendations
 - Equilibrium Worksheet Pasadena User Reviews and Ratings

- Equilibrium Worksheet Pasadena and Bestseller Lists
- 5. Accessing Equilibrium Worksheet Pasadena Free and Paid eBooks
 - Equilibrium Worksheet Pasadena Public Domain eBooks
 - Equilibrium Worksheet Pasadena eBook Subscription Services
 - Equilibrium Worksheet Pasadena Budget-Friendly Options
- 6. Navigating Equilibrium Worksheet Pasadena eBook Formats
 - ePub, PDF, MOBI, and More
 - Equilibrium Worksheet Pasadena Compatibility with Devices
 - Equilibrium Worksheet Pasadena Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Equilibrium Worksheet Pasadena
 - Highlighting and Note-Taking Equilibrium Worksheet Pasadena
 - Interactive Elements Equilibrium Worksheet Pasadena
- 8. Staying Engaged with Equilibrium Worksheet Pasadena
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Equilibrium Worksheet Pasadena
- 9. Balancing eBooks and Physical Books Equilibrium Worksheet Pasadena
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Equilibrium Worksheet Pasadena
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Equilibrium Worksheet Pasadena
 - Setting Reading Goals Equilibrium Worksheet Pasadena
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Equilibrium Worksheet Pasadena
 - Fact-Checking eBook Content of Equilibrium Worksheet Pasadena
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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berapa batas usia ideal hamil simak penelitian berikut ini - Oct 08 2022

1 menggunakan kalkulator kehamilan online kalkulator kehamilan online akan mudah untuk digunakan dan diakses oleh banyak orang untuk sekedar mengetahui usia kehamilan kamu

apa perbedaan kehamilan pada usia 20 30 dan 40 tahun - Aug 06 2022

dalam generasi yang solid dikatakan bahwa usia baik baik saja untuk kehamilan dan persalinan pada usia 20 tahun hingga 30 tahun meskipun kematian dalam persalinan dikatakan berada

bab ii tinjauan pustaka a usia kehamilan 1 klasifikasi - Nov 09 2022

may 26 2018 rp 79 500 boleh dibilang pada usia 30 35 tahun merupakan masa masa transisi kehamilan pada usia ini harus dijaga dengan baik pastikan kondisi tubuh dan

bab ii tinjauan pustaka 2 1 konsep dasar kehamilan - Jun 16 2023

buku kia hal 28 c hindari menyentuh mata hidung dan mulut dengan tangan yang belum dicuci d sebisa mungkin hindari kontak dengan orang yang sedang sakit pedoman bagi ibu

cara menghitung tabel usia kehamilan dan lingkaran perut - Dec 10 2022

fertilisasi sampai kelahiran bayi kehamilan normal biasanya berlangsung dalam waktu 40 minggu usia kehamilan tersebut dibagi menjadi 3 trimester yang masing masing

bab ii tinjauan pustaka 2 1 definisi berat bayi lahir - Mar 01 2022

sep 29 2022 berdasarkan 167 responden didapatkan hasil usia ibu hamil tidak beresiko 20 35 tahun sebanyak 101 orang 60 5 paritas 2 anak sebanyak 104 orang 62 3 jarak

kehamilan tanda pembuahan dan perkembangan - Jan 11 2023

usia 21 35 tahun adalah masa di mana ibu hamil memiliki risiko kesehatan paling rendah secara umum masa masa ini disebut sebagai waktu ideal untuk hamil dan melahirkan bagi

cara menghitung usia kehamilan dengan benar okadoc - Jul 05 2022

hidup di luar kandungan dengan berat badan kurang dari 1000 gram atau usia kehamilan kurang dari 28 minggu manuba 2010 dapat disimpulkan bahwa abortus adalah

bab ii tinjauan pustaka poltekkes bandung - Jul 17 2023

7 bab ii tinjauan pustaka 2 1 konsep dasar kehamilan 2 1 1 pengertian kehamilan kehamilan adalah suatu masa yang dimulai

dari konsepsi sampai lahirnya janin lamanya

bab ii tinjauan pustaka a konsep dasar kehamilan 1 - Jun 04 2022

1 bayi prematur preterm kurang bulan usia kehamilan kurang dari 37 minggu 259 hari 2 bayi cukup bulan aterm usia kehamilan antara 37 dan 42 minggu 259 293 hari 3 bayi

bab ii tinjauan pustaka 2 1 kehamilan 2 1 1 definisi - Sep 07 2022

hari 40 minggu dan tidak lebih dari 300 hari 43 minggu pembagian kehamilan dibagi dalam 3 trimester trimester i dimulai dari konsepsi sampai tiga bulan 0 12 minggu trimester ii