

Algebra 2 Unit 12: Statistical Notes

Variability and Sampling

Data is everywhere. In its raw form, pages, it's in our science classes, it shows up in economics, statistics and anywhere else that variability occurs. Variability is simply the property of outcomes being different. The tools of statistics are designed to explain this variability. There are many types of variability. It is good to understand these various tools in order to understand their strengths and weaknesses.

Example 1 The following types of variability can change outcomes of a data set. For each, give an example that may hold.

- (a) **Observational or observational variability:** Variability that is introduced due to either our measuring instruments not being precise enough or differences in how the different people read the measurements.

2 students measure time for a ball to roll down a ramp
→ different reaction times

- (b) **Natural variability or human-induced variability:** Variability that occurs due to the fact that members of populations are simply different.

2 people work out same amount of time
→ different people use different amounts of weight

- (c) **Random Variability:** This type of variability is the result of noise and error. It can be because our tools assigned our populations or samples to two or more treatment groups and then observing the variability between the groups.

2 groups taking a test, one group allowed to sleep 8 hours, other group only 4 hours → different tests

- (d) **Sample Variability:** This is the type of variability that occurs when we take multiple samples. There is population variability. There is going to still be difference due to the variability of the samples or errors.

Multiple samples of people surveyed to determine political candidate preference

Remember, through all of our work in this book, we are really trying to explain the variability of data within either a population or a sample and then using this to determine if the variability was the introduced to one of the factors chosen or the variability of the others.

Emathinstruction Answer Key For Unit 4

Camilla Rothe



Emathinstruction Answer Key For Unit 4:

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1. The book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Emathinstruction Answer Key For Unit 4
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 - Chapter 4: Emathinstruction Answer Key For Unit 4 in Specific Contexts
 - Chapter 5: Conclusion
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web ans the mole concept in chemistry is a fundamental concept that allows us to connect the mass of a substance to the
number of particles atoms molecules or ions it contains it is based on avogadro s number which states that one mole of any
substance contains 6 022 x 10²³ particles 2