

WK	LSN	TOPIC/S-TOPIC	OBJECTIVES	L/ACTIVITIES	L/T AIDS	REFERENCE	REMARKS
1	1		SCHOOL OPENING				
	2	Structure of the atom and the periodic table - Structure of the atom	By the end of the lesson, the learner should be able to:- - Define atom - Name the subatomic particles and Where found - Describe the structure of the atom and label it	- Notes taking - Discussion - Drawing and labeling the atom	Chart of the structure of the atom	KLB BK 2 Pg 1	
	3	Atomic number and Atomic mass	By the end of the lesson, the learner should be able to:- - Define atomic number and atomic mass - Do simple calculations on how to find either of the given one of them	- Discussion - Notes taking - Calculation	Past questions	KLB BK 2 Pg 2-3	
	4	Isotopes	By the end of the lesson, the learner should be able to:- - Define isotopes - Name some elements that exhibit Isotopes - Name isotopes of hydrogen	- Discussion - Notes taking - Calculation	- Chalkboard - Charts	KLB BK 2 Pg 4	
2	1&2	Energy levels and Electron arrangement	By the end of the lesson, the learner should be able to:- - Write the electronic arrangement of The 1 st 20 elements - State the importance of the number of energy levels as well as the number of electrons in the outermost energy level	- Discussion - Notes taking - Writing of the E.C	Charts of E.C of the 1 st 20 elements	KLB BK 2 Pg 4-7	
	3 4	Development of the Periodic table	By the end of the lesson, the learner should be able to:- - Build up the periodic table for the 1 st 20 elements based on the no. of energy levels and no. of electrons in the outermost energy level. - Explain the position of an element in The table	- Discussion - Notes taking - Class activity	- Manila - Plasticine - Ruler - Felt pens - Charts of the 1 st 20 elements	KLB BK 2 Pg 8-9	
3	1&2	Relative atomic mass	By the end of the lesson, the learner				

My Scheme Of Work For 3rd Term In Chemistry

Albert A Gayle



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Table of Contents My Scheme Of Work For 3rd Term In Chemistry

1. Understanding the eBook My Scheme Of Work For 3rd Term In Chemistry
 - The Rise of Digital Reading My Scheme Of Work For 3rd Term In Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying My Scheme Of Work For 3rd Term In Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an My Scheme Of Work For 3rd Term In Chemistry
 - User-Friendly Interface

4. Exploring eBook Recommendations from My Scheme Of Work For 3rd Term In Chemistry
 - Personalized Recommendations
 - My Scheme Of Work For 3rd Term In Chemistry User Reviews and Ratings
 - My Scheme Of Work For 3rd Term In Chemistry and Bestseller Lists
5. Accessing My Scheme Of Work For 3rd Term In Chemistry Free and Paid eBooks
 - My Scheme Of Work For 3rd Term In Chemistry Public Domain eBooks
 - My Scheme Of Work For 3rd Term In Chemistry eBook Subscription Services
 - My Scheme Of Work For 3rd Term In Chemistry Budget-Friendly Options
6. Navigating My Scheme Of Work For 3rd Term In Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - My Scheme Of Work For 3rd Term In Chemistry Compatibility with Devices
 - My Scheme Of Work For 3rd Term In Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of My Scheme Of Work For 3rd Term In Chemistry
 - Highlighting and Note-Taking My Scheme Of Work For 3rd Term In Chemistry
 - Interactive Elements My Scheme Of Work For 3rd Term In Chemistry
8. Staying Engaged with My Scheme Of Work For 3rd Term In Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers My Scheme Of Work For 3rd Term In Chemistry
9. Balancing eBooks and Physical Books My Scheme Of Work For 3rd Term In Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection My Scheme Of Work For 3rd Term In Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine My Scheme Of Work For 3rd Term In Chemistry
 - Setting Reading Goals My Scheme Of Work For 3rd Term In Chemistry
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of My Scheme Of Work For 3rd Term In Chemistry
 - Fact-Checking eBook Content of My Scheme Of Work For 3rd Term In Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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