Gold Penny Lab

Objective: Students will be also ring Metallis Banding and a ringle replacement reaction.

Macharine's:

- Zinc Powder Dinc can be rinsed with water and used again for this late.
- 3 M NorOH (Extremely Countic Googles Required)
- Baseliner with Wilsten
- Evaporating Dish
- Brumsen Burmen



Procedure:

- 1. Place a few groms of Zinc into an evaporating state.
- 2. Add obout 10mL of NoOM
- Slowly heat the solution, be coreful not to ball the solution.
- 4. Add a permy and siminer until the penny becomes silvery. It may be helpful to cover the penny in Zinc. (You may need to flip the penny to cover the sides equality.)
- 5. Use tongs to pick up the penny and drop it into a beaker of water to rince it.
- Hold one edge of the penny with tungs and flip it back and forth in the fire for a few seconds.
 Repeat until the penny turns gold.
- 7. Immediately drop it into the water again.

Responding.

- Freedict the products and incloses the equation Co + 2niOtti, ... 2n + CoOttis
- 2. What type of reaction occurred when you replaced the Copper with the Encil Single Replacement
- 3. What type of bonding was this due to! Defend your attends. Could defend remain bending blue to along the blue to place by the hydroxide, engages by bending bending blue to place to place by the business bending bending.
- 4. Name the ollay given to the two metals found on the penny. (You may need to look this up?) !!-----
- Permiss mode prior to 1982 are made from an other composed of 98% Capper and 2% Zinc.
 Why might permiss be made from an alloy rather than pure capper? Cost, strength, present rath.
- If you examined your penny microscopically, do you thing the penny would be homogeneous or festerogeneous throughout?

Gold Penny Lab Answers

Thirumalaisamy P. Velavancorresponding

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