

## Worksheet # 1 & Gram Formula Mass Answers

- |  |                  |
|--|------------------|
| 1. $4.5 \times 10^{23}$ mcls $\text{H}_2\text{O}$                      | 1. 158.04 g/mol  |
| 2. $3 \times 10^{24}$ atoms Fe   | 2. 74.55 g/mol   |
| 3. $5.73 \times 10^{-4}$ mol Cs  | 3. 142.05 g/mol  |
| 4. $1.1 \times 10^5$ mol Na  | 4. 164.10 g/mol  |
| 5. $2.1 \times 10^{23}$ mcls $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ | 5. 342.17 g/mol  |
| 6. $9.55 \times 10^{-2}$ mol Cu  | 6. 149.12 g/mol  |
| 7. $1.204 \times 10^{24}$ atoms Fr                                     | 7. 249.70 g/mol  |
| 8. 2.000 mol He  | 8. 262.87 g/mol  |
| 9. $5.6 \times 10^2$ mol S   | 9. 219.52 g/mol  |
|  | 10. 458.16 g/mol |

# Gram Formula Mass Worksheet Answers

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