

## Worksheet # 1 & Gram Formula Mass Answers

- |  |                  |
|--|------------------|
| 1. $4.5 \times 10^{23}$ mcls $\text{H}_2\text{O}$                      | 1. 158.04 g/mol  |
| 2. $3 \times 10^{24}$ atoms Fe   | 2. 74.55 g/mol   |
| 3. $5.73 \times 10^{-4}$ mol Cs  | 3. 142.05 g/mol  |
| 4. $1.1 \times 10^5$ mol Na  | 4. 164.10 g/mol  |
| 5. $2.1 \times 10^{23}$ mcls $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ | 5. 342.17 g/mol  |
| 6. $9.55 \times 10^{-2}$ mol Cu  | 6. 149.12 g/mol  |
| 7. $1.204 \times 10^{24}$ atoms Fr                                     | 7. 249.70 g/mol  |
| 8. 2.000 mol He  | 8. 262.87 g/mol  |
| 9. $5.6 \times 10^2$ mol S   | 9. 219.52 g/mol  |
|  | 10. 458.16 g/mol |

# Gram Formula Mass Answers

**Chandan Sengupta**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

## **Gram Formula Mass Answers:**

**Regents Exams and Answers: Chemistry--Physical Setting Revised Edition** Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day **E3**

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