

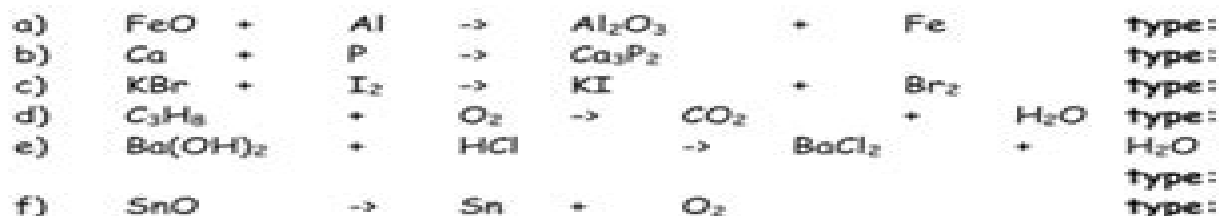
## Grade 10 Chemistry Review

1. Use the periodic table to answer the following questions concerning calcium.
  - a) In what period is it located?
  - b) In what group is it located?
  - c) What is its atomic number?
  - d) What is its atomic mass?
  - e) In what state would it be found at room temperature?
2. Explain why O forms a -2 charged ion. Provide a written explanation and Bohr-Rutherford diagrams for both the atom and the ion.
3. Write formulas for each of the following compounds.

sulfur pentafluoride  
calcium oxide  
nitrogen gas  
iron (III) sulfide

sodium sulfate  
tin(II) bromide  
calcium phosphate  
ammonia

4. Name the type of reaction and balance each equation.



5. Write word equations for each of the reactions in #4.  $\text{C}_3\text{H}_8$  is propane.
6. Describe a standard test for each of the following substances:
  - a) oxygen gas
  - b) hydrogen gas
  - c) carbon dioxide gas
  - d) water vapour

# Grade 10 Science Chemistry Unit Test Cscope

**Ken Dobson**



## **Grade 10 Science Chemistry Unit Test Cscope:**

Co-ordinated Science Ken Dobson,1993      **Explaining Chemical Processes** Jim (James William) Ross,Jim Ross,Mike Lattner,2004 An introduction to the four simplest types of chemical reactions this unit begins with decomposition and synthesis reactions and moves to the reactions of acids and bases The relationship between atomic structure chemical behavior and organization of the periodic table is extended from the Grade Nine Chemistry unit The unit is composed of three major sections 1 Decomposition and synthesis reactions with their one to many and many to one structure are the easiest to work with Within this context students will learn how to represent chemical change four different parallel ways diagrams symbolic balanced equations word equations and mass equations We clearly intend this unit to anticipate and support the introduction of the mole concept in Grade 11 2 Metals non metals and pH The origins of acid and base behaviour are probed here These reactions exemplify single displacement reactions 3 Reactions of Acids and Bases In both metal acid reactions and acid base reactions we continue to use the four representations These reactions exemplify the double displacement reaction type Emphasis is placed on naming the chemicals involved At the end of each section is a thorough quiz

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