

Name \_\_\_\_\_  
stoichiometry

Test next class over

1. In the formation of carbon dioxide from carbon monoxide and oxygen, how many moles of carbon monoxide are needed to react completely with 7.0 moles of oxygen gas?
2. How many moles of oxygen are required to burn 45.8 grams of ethane gas,  $C_2H_6$  at standard conditions?
3. How many grams of oxygen are produced by the decomposition of 55.0 grams of potassium chlorate,  $KClO_3$ ?
4. A chemist performs the synthesis of sodium chloride from its elements. The chemist begins with 46.55 grams of sodium. How many grams of chlorine are needed?
5. How many grams of water can be prepared from 5.68 moles of hydrogen at standard conditions?
6. Suppose that an excess of propane,  $C_3H_8$  burns in 62.0 g of  $O_2$ . How many moles of  $H_2O$  will be formed?

# Grade 1 stoichiometry Practice Test

**David Kirk**



## **Grade 1 stoichiometry Practice Test:**

## Embracing the Beat of Term: An Mental Symphony within **Grade 1stoichiometry Practice Test**

In a world taken by displays and the ceaseless chatter of instant communication, the melodic splendor and emotional symphony produced by the prepared term usually diminish into the back ground, eclipsed by the relentless sound and interruptions that permeate our lives. Nevertheless, set within the pages of **Grade 1stoichiometry Practice Test** a marvelous literary treasure filled with organic thoughts, lies an immersive symphony waiting to be embraced. Crafted by an elegant musician of language, this captivating masterpiece conducts viewers on an emotional trip, skillfully unraveling the hidden melodies and profound influence resonating within each carefully crafted phrase. Within the depths of the moving evaluation, we will investigate the book is central harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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