

Molar Concentration Worksheet

Name: _____

Complete all questions including units in your calculations. Each question is worth 4 marks.

You may use Google to find the molar masses (ex. type 'molar mass HCl = ' and it will give you the answer).

1. Every 5.5 L of battery acid contains 490.0 g of sulfuric acid, H_2SO_4 . Calculate the concentration of sulfuric acid in the battery acid in mol/L.
2. Lily added 35 grams of sodium carbonate to a volume of 3,400 mL. What is the concentration of her solution in mol/L? (answer 0.097 mol/L)
3. If Jaclyn adds water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity (concentration in mol/L) of the diluted solution be?
4. Explain how Savannah could make 450 mL of a 0.25 M calcium chloride solution in the lab from solid calcium chloride if the only volumetric flask she has is a 500 mL volumetric flask. Show all calculations.

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SA Dillow



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