

Quarter 3 Exam Review

Multiple Choice

Always, the choice that best completes the statement or answers the question.

- _____ 1. In a chemical reaction
- the mass of the reactants equals the mass of the products.
 - the mass of the product is greater than the mass of reactants.
 - the material atoms in the reactants and products must change.
 - energy is lost must be added to the reactants.
- _____ 2. When a ratio, provided by a chemical reaction, compares facts the solution it is called
- a percent.
 - a percent.
 - a percent.
 - the mass of the product.
- _____ 3. In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is
- H
 - H₂
 - H₂
 - H₂
- _____ 4. To balance a chemical equation, it may be necessary to adjust the
- coefficients.
 - formulas of the products.
 - subscripts.
 - reactant of products.
- _____ 5. According to the law of conservation of mass, the total mass of the reacting substances is
- always more than the total mass of the products.
 - always less than the total mass of the products.
 - sometimes more and sometimes less than the total mass of the products.
 - always equal to the total mass of the products.
- _____ 6. In the chemical equation $2\text{Mg}(s) + \text{O}_2(g) \rightarrow 2\text{MgO}(s)$
- Mg represents the product magnesium.
 - O₂ represents the reactant oxygen.
 - the reaction yield magnesium.
 - O₂ represents the product oxygen gas.
- _____ 7. When the reaction $2\text{FeCl}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 6\text{HCl}$ is correctly balanced, what is the coefficient of Fe?
- 4
 - 2
 - 3
 - 6
- _____ 8. Which coefficients correctly balance the formula equation $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$?
- 1, 1, 2
 - 1, 2, 1
 - 1, 1, 1
 - 1, 2, 1
- _____ 9. In which kind of reaction do two or more substances combine to form a new compound?
- decomposition reaction
 - double displacement reaction
 - single displacement reaction
 - synthesis reaction
- _____ 10. The equation $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$ is the general equation for a
- synthesis reaction.
 - decomposition reaction.
 - single displacement reaction.
 - double displacement reaction.

General Chemistry Review Exam Guide Answers Packet

JA Banks



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Table of Contents General Chemistry Review Exam Guide Answers Packet

1. Understanding the eBook General Chemistry Review Exam Guide Answers Packet
 - The Rise of Digital Reading General Chemistry Review Exam Guide Answers Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Review Exam Guide Answers Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry Review Exam Guide Answers Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Review Exam Guide Answers Packet
 - Personalized Recommendations
 - General Chemistry Review Exam Guide Answers Packet User Reviews and Ratings
 - General Chemistry Review Exam Guide Answers Packet and Bestseller Lists

5. Accessing General Chemistry Review Exam Guide Answers Packet Free and Paid eBooks
 - General Chemistry Review Exam Guide Answers Packet Public Domain eBooks
 - General Chemistry Review Exam Guide Answers Packet eBook Subscription Services
 - General Chemistry Review Exam Guide Answers Packet Budget-Friendly Options
6. Navigating General Chemistry Review Exam Guide Answers Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - General Chemistry Review Exam Guide Answers Packet Compatibility with Devices
 - General Chemistry Review Exam Guide Answers Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry Review Exam Guide Answers Packet
 - Highlighting and Note-Taking General Chemistry Review Exam Guide Answers Packet
 - Interactive Elements General Chemistry Review Exam Guide Answers Packet
8. Staying Engaged with General Chemistry Review Exam Guide Answers Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry Review Exam Guide Answers Packet
9. Balancing eBooks and Physical Books General Chemistry Review Exam Guide Answers Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry Review Exam Guide Answers Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine General Chemistry Review Exam Guide Answers Packet
 - Setting Reading Goals General Chemistry Review Exam Guide Answers Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of General Chemistry Review Exam Guide Answers Packet
 - Fact-Checking eBook Content of General Chemistry Review Exam Guide Answers Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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