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$\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{NH}_4^+ + \text{SO}_4^{2-}$

$\text{C}_5\text{H}_9\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

$\text{Pb} + \text{H}_3\text{PO}_4 \rightarrow \text{H}_2 + \text{Pb}_3\text{PO}_4$

$2\text{Li}_3\text{N} + \text{NH}_4\text{NO}_3 \rightarrow \text{LiNO}_3 + \text{NH}_4\text{OH}$

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web 1 3nabr 1h3po4 1na3po4 3hbr type of reaction double displacement 2 3ca oh 2 1al2 so4 33caso4 2al oh 3 type of reaction double displacement 3 3mg 1fe2o3 2fe 3mgo type of reaction single displacement 4 1c2h4 3o2 2co2 2h2o type of reaction combustion 5 2pbso4 2pbso3 1o2 type of reaction

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web a chemical reaction is a process in which one or more substances are converted to one or more different substances the starting substances are called the reactants and the new substances that form are called the products

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web study guide chemical reactions 1 give an example of a chemical reaction leaves turning color fireworks food spoiling 2 differentiate between a physical and chemical change physical does not change the identity of the matter can be changed back easily chemical new substance is created with new physical and chemical