

Assignment #2 – Compound Names and Formulas

Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Formula Of Ionic Compound Lab Answer Key

**J. Theo Kloprogge, Concepcion P.
Ponce, Tom Loomis**



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Table of Contents Formula Of Ionic Compound Lab Answer Key

1. Understanding the eBook Formula Of Ionic Compound Lab Answer Key
 - The Rise of Digital Reading Formula Of Ionic Compound Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Formula Of Ionic Compound Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Formula Of Ionic Compound Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Formula Of Ionic Compound Lab Answer Key
 - Personalized Recommendations
 - Formula Of Ionic Compound Lab Answer Key User Reviews and Ratings

- Formula Of Ionic Compound Lab Answer Key and Bestseller Lists
- 5. Accessing Formula Of Ionic Compound Lab Answer Key Free and Paid eBooks
 - Formula Of Ionic Compound Lab Answer Key Public Domain eBooks
 - Formula Of Ionic Compound Lab Answer Key eBook Subscription Services
 - Formula Of Ionic Compound Lab Answer Key Budget-Friendly Options
- 6. Navigating Formula Of Ionic Compound Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Formula Of Ionic Compound Lab Answer Key Compatibility with Devices
 - Formula Of Ionic Compound Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Formula Of Ionic Compound Lab Answer Key
 - Highlighting and Note-Taking Formula Of Ionic Compound Lab Answer Key
 - Interactive Elements Formula Of Ionic Compound Lab Answer Key
- 8. Staying Engaged with Formula Of Ionic Compound Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Formula Of Ionic Compound Lab Answer Key
- 9. Balancing eBooks and Physical Books Formula Of Ionic Compound Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Formula Of Ionic Compound Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Formula Of Ionic Compound Lab Answer Key
 - Setting Reading Goals Formula Of Ionic Compound Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Formula Of Ionic Compound Lab Answer Key
 - Fact-Checking eBook Content of Formula Of Ionic Compound Lab Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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