

It's Elementary — March Madness



March Madness is upon us! Your students are probably already working on their next assignment in bracketology, predicting the field for the NCAA Basketball Tournament. (Although the word doesn't appear in Webster's, there is an article on bracketology on Wikipedia, so it must be real.)

Let's apply the principles of bracketology to the properties of the elements. The competition among the 64 elements is stiff, and there is a wealth of information for students to research, in this "elementary" version of March Madness.

Review of Element Properties

Define each of the following properties of the elements and briefly describe any periodic trend in the property across a row or down a column in the periodic table.

- Ionization energy
- Atomic radius

Tournament Rules

Here are the rules for predicting the winners in each round of *It's Elementary—March Madness*. In the event of a tie in the properties of two competing elements in any round of the tournament, the element with the larger atomic mass always wins.

- First round: Research the date of discovery of each element. In each bracket, *the element that was discovered earlier (in its free element form) wins* and proceeds to the second round. If an element has been known since ancient times, assign it a discovery date of zero.
- Second round: Compare the ionization energy of the elements in each bracket. *The element with the higher ionization energy is the winner* and advances to the Sweet 16.
- Third round (Sweet 16): Compare the group numbers of the elements—*the winner is the element with the larger group number using the international (IUPAC) system (Groups 1–18)*.
- Fourth round: *The element with the larger atomic radius wins this round and earns a trip to the Final Four.* Use the atomic or covalent radius only, not the van der Waals radius.
- Semifinals (Final Four): *Solve the following riddles to determine the two elements that will compete for the championship.*
 - ❖ This "salt-maker" is also a rainmaker when its silver salt is scattered into clouds.
 - ❖ Once a sedative and cure for nervous tension, the ion of this element is now a mite or commonplace expression.
- Finals: It's often said that there's no _____ in teams, but it is the winner in this field!

NGSS Alignment

This laboratory activity relates to the following Next Generation Science Standards (2013):

Disciplinary Core Ideas: Middle School
MS-PS1 Matter and Its Interactions
PS1.A: Structure and Properties of Matter

Disciplinary Core Ideas: High School
HS-PS1 Matter and Its Interactions
PS1.A: Structure and Properties of Matter

Science and Engineering Practices
Asking questions and defining problems
Analyzing and interpreting data

Crosscutting Concepts
Patterns
Structure and function

References

John Emsley, *Nature's Building Blocks*, Oxford University Press, Oxford, England (2001).
WebElements' "Periodic Table," <http://www.weblelements.com> (accessed February 2008).
Flinn Scientific Periodic Table of the Elements, Catalog No. AP9020.

Flinn Scientific March Madness Answers

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Flinn Scientific March Madness Answers:

Unveiling the Magic of Words: A Report on "**Flinn Scientific March Madness Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Flinn Scientific March Madness Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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Table of Contents Flinn Scientific March Madness Answers

1. Understanding the eBook Flinn Scientific March Madness Answers
 - The Rise of Digital Reading Flinn Scientific March Madness Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Flinn Scientific March Madness Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Flinn Scientific March Madness Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Flinn Scientific March Madness Answers
 - Personalized Recommendations
 - Flinn Scientific March Madness Answers User Reviews and Ratings

- Flinn Scientific March Madness Answers and Bestseller Lists
- 5. Accessing Flinn Scientific March Madness Answers Free and Paid eBooks
 - Flinn Scientific March Madness Answers Public Domain eBooks
 - Flinn Scientific March Madness Answers eBook Subscription Services
 - Flinn Scientific March Madness Answers Budget-Friendly Options
- 6. Navigating Flinn Scientific March Madness Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Flinn Scientific March Madness Answers Compatibility with Devices
 - Flinn Scientific March Madness Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Flinn Scientific March Madness Answers
 - Highlighting and Note-Taking Flinn Scientific March Madness Answers
 - Interactive Elements Flinn Scientific March Madness Answers
- 8. Staying Engaged with Flinn Scientific March Madness Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Flinn Scientific March Madness Answers
- 9. Balancing eBooks and Physical Books Flinn Scientific March Madness Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Flinn Scientific March Madness Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Flinn Scientific March Madness Answers
 - Setting Reading Goals Flinn Scientific March Madness Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Flinn Scientific March Madness Answers
 - Fact-Checking eBook Content of Flinn Scientific March Madness Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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