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**A comprehensive guide to
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**Simple, logical and colour-coded notes
with step-by-step examples**

APPROVED on the DBE National Catalogue

**COVERS THE FUNDAMENTALS OF GRADE 12 MATHEMATICS
AND COMPLEMENTS THE MATERIAL IN ANY CLASS TEXT**

GRADE 12

KEVIN SMITH

SUITABLE FOR IEB AND NATIONAL CURRICULUM STUDENTS

Mathematics Grade 12 Study Guide

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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$$\text{I}_2\text{O}_5 + \text{CO} \rightarrow \text{I}_2 + \text{CO}_2$$

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$$2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$$

$$\text{Sr} \rightarrow \text{Sr}^{2+} \text{ oxidized}$$

$$\text{O} \rightarrow \text{O}^{2-} \text{ reduced}$$

$$2\text{Li} + \text{S} \rightarrow 2\text{Li}_2\text{S}$$

$$\text{Li} \rightarrow \text{Li}^+ \text{ oxidized}$$

$$\text{S} \rightarrow \text{S}^{2-} \text{ reduced}$$

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