STOICHIOMETRY: MOLE-MOLE PROBLEMS

Name _____

1. N₂ + 3H₂ → 2NH₃

How many moles of hydrogen are needed to completely react with two moles of nitrogen?

6 moles Hz

2. 2KCIO, → 2KCI + 3O,

How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

9 moles of Oz

3. Zn + 2HCl → ZnCl, + H,

How many moles of hydrogen are produced from the reaction of three moles of zinc with an excess of hydrochlaric acid?

3 moles of Hz

4. C,H, + 50, - 3CO, + 4H,O

How many moles of oxygen are necessary to react completely with four moles of propane (C,H₂)?

20 mol of Q2

5. K,PO, + AI(NO,), - 3KNO, + AIPO,

How many moles of potassium nitrate are produced when two moles of potassium phosphate react with two moles of diuminum nitrate?

6 moles of KNO3

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SA Adler

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