Introduction to Combustion Concepts and Applications 3rd Edition Turns Solutions Manual

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PROBLEM 2-37

GIVEN: A system of fixed mass at constant pressure and temperature

FIND: Equivalent system form of Egtn. 2.35 (Find Law) which is used to define the heat of reaction.

Assumptions: expansion/contraction of the system boundary is a reversible process, only work done by system is boundary expansion/contraction

APPROACH: Write the first law for the system. Substitute the appropriate expression for constant pressure work and solve for the specific heat transfer.

sketch of system :



first law for a system :

for a system at constant pressure the reversible work can be expressed as

substituting into the first law

$$Q_2 - mP(v_1 - v_1) = m(u_2 - u_1)$$

 $Q_2 = m[(u_2 + P_{v_2}) - (u_1 + P_{v_1})] = m(h_2 - h_1)$
 $Q_2 = Q_2/m = h_2 - h_1$

if state 1 is the readonts and state 2 is the products then $q = h_p - h_e$ same form as Egth 2.35

Comments: Note that the first law for a constant pressure system has the same form as for a control volume. This is only true for constant pressure Also, the internal energies u, and uz are mosture internal energies. Even assuming ideal gas behavior, u, x uz at constant temperature if the composition changes. (see Jades experiment with gases and internal energy)

Introduction To Combustion Turns Not Solution Manual

Joacim Rocklöv

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