

# AGRICULTURAL SCIENCES

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GRADE 11

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NOTES AND ACTIVITY BOOKLET



By E.C. Mafuza

# Grade 11 Exemplar Agricultural Science

**SJ Ball**



## **Grade 11 Exemplar Agricultural Science:**

## Decoding **Grade 11 Exemplar Agricultural Science**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Grade 11 Exemplar Agricultural Science**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$   
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$   
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$   
 $6\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$   
 $3\text{K}_2\text{SO}_4 + 6\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{H}_2\text{O} \rightarrow 4\text{H}_2\text{O} + 2\text{CO}_2 + 2\text{H}_2\text{O} + 8\text{H}_2\text{O} + 1\text{C}_3\text{H}_8 + 5\text{O}_2 + 3\text{CO}_2 + 4\text{H}_2\text{O}$

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 $1\text{N}_2 + \text{F}_2 \rightarrow \text{N}_2\text{F}_4$   
 $2\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow 3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{NaBr} + \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$  using the equation from problem 2 above answer the following

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 $2\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + 1\text{KHCO}_3 + 1\text{H}_2\text{O} + 1\text{KBr} + \text{CO}_2 + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + 1\text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SnO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + 1\text{N}_2\text{O}_3$  using the following equation  $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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web may 28 2020 write the balanced equation for the reaction and calculate how much  $\text{SiO}_2$  is required to produce 3.00 kg of  $\text{SiC}$  answer  $\text{Ce} + \text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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