

- d. Secondary passive transport protein
10. An intermediate compound X is formed during the production of urea through *ammonia/carbon dioxide* technology. What is the molecular formula of X?
- $\text{NH}_2\text{COONH}_2$
  - $\text{NH}_2\text{COONH}_4$
  - $\text{NH}_4\text{COONH}_2$
  - $\text{NH}_4\text{COONH}_4$
11. Which of the following are recycled in the manufacture of sodium Carbonate by Solvay's process?
- $\text{CO}_2$  and  $\text{NH}_4\text{Cl}$
  - $\text{CO}_2$  and  $\text{NH}_3$
  - $\text{NaCl}$  and  $\text{CaO}$
  - $\text{NaCl}$  and  $\text{NH}_3$

### Group 'B'

Give short answer to the following questions.

(8 × 5=40)

- I. An element X has 2 electrons in K shell, 8 electrons in L shell and 5 electrons in M shell.
- Identify the element X and write the number of protons and electrons in it. [3]
  - Size of  $\text{X}^{3-}$  ion is greater than that of X atom though both contain the same number of protons. Give reason [1]
  - Write down the formula of one of the compounds of X where X is in -3 oxidation state. [1]

OR

Know-how about ionization energy (IE) of elements is crucial aspect in the study of chemical bonding whether they form ionic or covalent bond. The first ionization energies of period second elements are given below

|                           |  |  |  |  |  |  |  |  |
|---------------------------|--|--|--|--|--|--|--|--|
| of elements               |  |  |  |  |  |  |  |  |
| ionization energy(kJ/mol) |  |  |  |  |  |  |  |  |

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