



NATIONAL  
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2  
(CHEMISTRY)  
EXEMPLAR

Duration: 120  
Time: 1 hour



The question paper consists of 12 pages, including 4 extra sheets.  
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## QUESTION 8

### PHYSICAL SCIENCES PAPER 2

### ACID AND BASES

**GRADE 11**

**NOVEMBER 2020**

QUESTION 8: ACID AND BASES (10)

8.1. Draw a potential energy diagram for the reaction represented by

The following equation shown on the page:

a. Heat of this reaction (20)  
b. Activation energy (10)

QUESTION 8 (Start on a NEW page)

8.2. Sodium carbonate is an ionic compound with the following percentage composition:

Na 27.0%, C 14.3%, O 58.7%

The molar mass of the compound is 106 g mol<sup>-1</sup>

8.2.1. Calculate the mass percentage of sodium (10)  
8.2.2. Determine the number of moles of sodium carbonate (10)  
8.2.3. Determine the volume of sodium carbonate (10)

8.3. Concentrated sulphuric acid is a mixture of sulphuric acid and water. It is a mixture of sulphuric acid and water. It is a mixture of sulphuric acid and water.

8.3.1. Calculate the concentration of the sulphuric acid (10)  
8.3.2. Calculate the concentration of the sulphuric acid (10)

8.4. Sodium carbonate reacts with hydrochloric acid to form sodium chloride and carbon dioxide.

$\text{Na}_2\text{CO}_3 + 2\text{HCl} \rightarrow 2\text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$

Calculate the volume of carbon dioxide gas that would be produced at STP if 10 g of sodium carbonate reacts completely. (10)

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