

INSTRUCTIONS AND INFORMATION

1. This question paper consists of THREE SECTIONS.

SECTION A: 25 Marks (Compulsory)

SECTION B: 25 Marks (Compulsory)

SECTION C: 20 Marks (Answer any TWO questions)

2. Answer in blue or black ink.
3. Start each section on a new page.
4. Read all the instructions carefully.
5. Number the questions as they appear in the question paper.
6. Leave a line open between each answer.
7. Write neatly and legibly.

SECTION A: (COMPULSORY)

Answer ALL the questions in this section.

QUESTION 1

Various options are provided as possible answers to the following questions. Choose the answer and write only the letter (A - D) next to the question number (e.g. 1.1 - 1.10) in the ANSWER BOOK, for example 1.4 B.

- 1.1 A volunteer can be defined as...
A. a person who believes that we and others are unjustified.
B. a person who is happy and enthusiastic.
C. a person who dislikes or fears people from other countries.
D. a person that gives their time and/or things to help an organisation. (7)
- 1.2 What guides people in their career choices?
A. Being in denial
B. Interests and abilities
C. Being depressed
D. Both A and C are correct (7)
- 1.3 Line dancing...
A. people standing in a circle and dancing
B. where people dance in rows
C. where people repeat a sequence of steps
D. both B and C. (7)
- 1.4 The National Student Financial Aid Scheme (NSFAS)...
A. is a requirement for university entry
B. is awarded to merit students with outstanding talent and achievement in sports
C. provides study loans to financially disadvantaged students
D. offers study loans at a very high interest rate (7)
- 1.5 At the end of the year you will have completed the...
A. SET Phase
B. GET Phase
C. Intermediate Phase
D. None of the above (7)

(26)

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SJ Ball



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of neutralization is thus absent brønsted lowry acid base behavior is formally independent of any solvent making it more all encompassing than the arrhenius model

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jan 30 2023 acetic acid is a weak acid $K_a = 1.8 \times 10^{-5}$ and acetate is a weak base $K_b = 5.6 \times 10^{-10}$ like acids strong and weak bases are classified by the extent of their ionization strong bases disassociate almost or entirely to completion in aqueous solution

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acids bases and enzymes many acids and bases in living things provide the pH that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper pH for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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Oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low pH value i.e. below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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Arrhenius's definition of acids and bases the earliest definition of acids and bases is Arrhenius's definition which states that

an acid is a substance that forms hydrogen ions H^+ when dissolved in water and a base is a substance that forms hydroxide ions OH^- when dissolved in water

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mar 24 2021 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is

$$acid + base \rightarrow water + salt$$

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