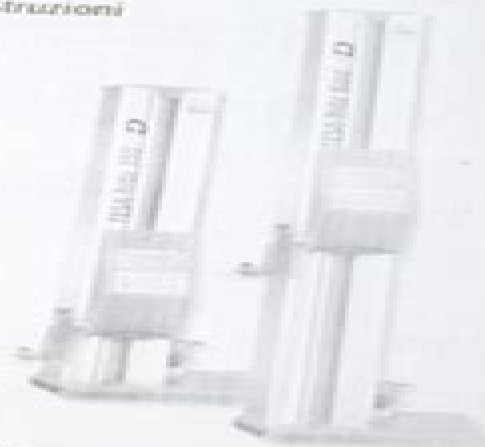


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# Manual For Tesa M600 Height Gauge

**Michael Seilmaier**



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web solution 1 the two half reactions balanced as if in acidic solution  $2\text{CN}^- + \text{Au} \rightarrow \text{AuCN} + 2\text{e}^-$   $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$  make electrons equal convert to basic solution  $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{AuCN} + 2\text{e}^-$  multiplied by a factor of 2  $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$  the final answer

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reactions  $6e^-$   $14h^+$   $Cr_2O_7^{2-}$   $2Cr^{3+}$   $7H_2O$   $Fe^{2+}$   $Fe^{3+}$   $e^-$  2 equalize the electrons  $6e^-$   $14h^+$   $Cr_2O_7^{2-}$   $2Cr^{3+}$   $7H_2O$   $6Fe^{2+}$   $6Fe^{3+}$   $6e^-$  multiplied by a factor of 6

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively  $C_6H_8O_6$   $2H_2NO_2$   $C_6H_6O_6$   $2H_2O$   $2NO$  choose 1 answer  $C_6H_8O_6$  and  $NO_2$  a  $C_6H_8O_6$  and  $NO_2$  no 2 and  $C_6H_8O_6$  b  $NO_2$  and  $C_6H_8O_6$  h and  $NO_2$  c h and

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