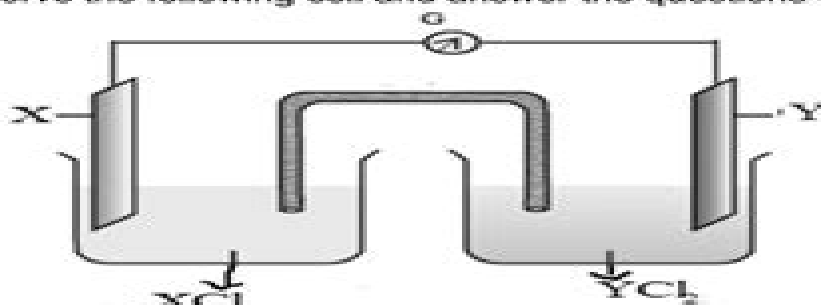


- 19 Observe the following cell and answer the questions that follow:



- Represent the cell shown in the figure.
- Name the carriers of the current in the salt bridge
- Write the reaction taking place at the anode.

1
½
½

(for visually challenged learners)

For the cell represented as:



- Identify the cathode and the anode
- Write the overall reaction

1
1

- 20 Complete the following reactions by writing the major and minor product in each case (any 2)



1



1



1

- 21 The presence of Carbonyl group in glucose is confirmed by its reaction with hydroxylamine. Identify the type of carbonyl group present and its position. Give a chemical reaction in support of your answer.

1
1

SECTION C

This section contains 7 questions with internal choice in one question. The following questions are short answer type and carry 3 marks each.

- 22 a. Write down the reaction occurring on two inert electrodes when electrolysis of copper chloride is done. What will happen if a concentrated solution of copper sulphate is replaced with copper chloride?

2

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Robert R. Redfield



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