#### Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

- 1. Metric units for mass, length, volume
- Metric to metric conversions (K H Da m/l/g D C M)
- Know how to put numbers in scientific notation and take them out.
- 4. Significant Digit rules
- Difference between chemical and physical properties and changes. There will be multiple questions on this.
- 6. Ways to tell if a chemical change (chemical reaction) is taking place.
- 7. Pure substances and mixtures
  - a. heterogeneous mixture
  - b. homogeneous mixture
- 8. Density & how it relates to floating and sinking
- Use the equation for density
- Density = Mass/volume (g/ml) (g) (ml)
- Know the phases of matter and the characteristics of each phase.
- 11. Law of Conservation of Matter
- 12. What is a mole in chemistry?
- 13. Convert grams to moles, and moles to grams
- Define an atom
- 15. Structure of an atom and where each subatomic particle is located.
  - a. Nucleus (contains protons and neutrons)
  - b. Protons (positive)
  - c. Neutrons (neutral)
  - d. Electrons (negative)
- 16. Isotopes
- 17. Elements
- 18. If given atomic # and mass, find the number of protons and neutrons in an atom.
- 19. Find the number of valence electrons, and tell what it says about reactivity.
- 20. Periodic Table
  - a. Know about the families
  - b. Group vs. period
- 21. Know about the properties of metals and nonmetals.
  - a. Ductile, malleable, conductors
- 22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons) Characteristics of Each

- Be able to name ionic compounds
- 24. Predict oxidation numbers
- 25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)

26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

# **How To Study For The Tn Chemistry Eoc Exam**

**Chao Zhang** 

#### **How To Study For The Tn Chemistry Eoc Exam:**

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Texas EOC Chemistry - the All-In-one Guide Mary Amboji, 2013-03-19 The complete guide for Texas EOC Chemistry has hundreds of guided and independent practice questions based on the categories tested on the EOC The dual coded questions help students gain mastery over scientific process skills Five category based tests are used to further hone your skills Calculator screen shots help you learn how to maximize your efficiency to solve complex problems There are two full length tests to prepare you for the actual exam This book can be used to prep students on chemistry 

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