



Delhi Public School, Prayagraj

HALF YEARLY EXAMINATION (2022-2023)

SUBJECT: - CHEMISTRY

CLASS: - XI

Duration: 3 Hours

Maximum Marks: 70

Name: - ~~XXXXXXXXXX~~

Class/Sec.: ~~XI/A~~

Instructions:

Q.No. 1 to 10 carries 1 mark each.

Q.No. 11 to 19 carries 2 marks each

Q.No. 20 to 28 carries 3 marks each

Q.No. 29 to 31 carries 5 marks each.

- Which of the following compounds has hydrogen with oxidation number +1?
(a) HBr (b) NaOH (c) LiH (d) H₂
- In the compounds KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$, the highest oxidation state is of the element:
(a) Mn (b) K (c) Cr (d) O
- If travelling at same speeds, which of the following matter waves have shortest wavelength?
(a) Proton (b) Neutron (c) Alpha particle (d) Electron
- Which of the following pairs of gases contains the same number of molecules?
(a) 16 gm of O_2 and 14 gm of N_2 (b) 8 gm of O_2 and 22 gm of CO_2
(c) 28 gm of N_2 and 22 gm of CO_2 (d) 32 gm of O_2 and 32 gm of N_2
- The number of water molecules present in a drop of water (volume 0.0018 ml) at room temperature is:
(a) 6.023×10^{23} (b) 1.084×10^{24} (c) 4.84×10^{23} (d) 6.023×10^{23}
- Which of the following is true?
(a) According to Pauli's exclusion principle, no two electrons in an atom can have the same values of quantum numbers n, l and m.
(b) The total energy of an electron in an orbital is half of its potential energy.
(c) The speed of an electron in an orbit increases with increase of quantum number n.

Grade 1chemistry Half Yearly Examination

Harvard University



Grade 1chemistry Half Yearly Examination:

Official Register of the Officers and Cadets of the U.S. Military Academy United States Military Academy,1910

Official Register of the Officers and Cadets United States Military Academy,1944 *Parliamentary Papers* Great Britain.
Parliament. House of Commons,1906 *Catalogue - Harvard University* Harvard University,1922 *Official Register*
Harvard University,1914 General Report - Transvaal Education Department Transvaal (South Africa). Education
Dept,1926 Pettigrew's New England Professional Directory 1904 Richard Richardson Pettigrew,1904 *Employment*
Outlook for Technicians Howard Rosen,1953 Journal of the American Medical Association ,1913 Includes proceedings of
the association papers read at the annual sessions and lists of current medical literature Bulletin of the United States
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This book delves into Grade 1chemistry Half Yearly Examination. Grade 1chemistry Half Yearly Examination is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Grade 1chemistry Half Yearly Examination, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Grade 1chemistry Half Yearly Examination
- Chapter 2: Essential Elements of Grade 1chemistry Half Yearly Examination
- Chapter 3: Grade 1chemistry Half Yearly Examination in Everyday Life
- Chapter 4: Grade 1chemistry Half Yearly Examination in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Grade 1chemistry Half Yearly Examination. The first chapter will explore what Grade 1chemistry Half Yearly Examination is, why Grade 1chemistry Half Yearly Examination is vital, and how to effectively learn about Grade 1chemistry Half Yearly Examination.

3. In chapter 2, the author will delve into the foundational concepts of Grade 1chemistry Half Yearly Examination. This chapter will elucidate the essential principles that must be understood to grasp Grade 1chemistry Half Yearly Examination in its entirety.

4. In chapter 3, the author will examine the practical applications of Grade 1chemistry Half Yearly Examination in daily life. This chapter will showcase real-world examples of how Grade 1chemistry Half Yearly Examination can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Grade 1chemistry Half Yearly Examination in specific contexts. This chapter will explore how Grade 1chemistry Half Yearly Examination is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Grade 1chemistry Half Yearly Examination. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Grade 1chemistry Half Yearly Examination.

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Table of Contents Grade 1chemistry Half Yearly Examination

1. Understanding the eBook Grade 1chemistry Half Yearly Examination
 - The Rise of Digital Reading Grade 1chemistry Half Yearly Examination
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 1chemistry Half Yearly Examination
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 1chemistry Half Yearly Examination
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 1chemistry Half Yearly Examination
 - Personalized Recommendations
 - Grade 1chemistry Half Yearly Examination User Reviews and Ratings
 - Grade 1chemistry Half Yearly Examination and Bestseller Lists
5. Accessing Grade 1chemistry Half Yearly Examination Free and Paid eBooks
 - Grade 1chemistry Half Yearly Examination Public Domain eBooks
 - Grade 1chemistry Half Yearly Examination eBook Subscription Services
 - Grade 1chemistry Half Yearly Examination Budget-Friendly Options
6. Navigating Grade 1chemistry Half Yearly Examination eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 1chemistry Half Yearly Examination Compatibility with Devices
 - Grade 1chemistry Half Yearly Examination Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 1chemistry Half Yearly Examination
 - Highlighting and Note-Taking Grade 1chemistry Half Yearly Examination
 - Interactive Elements Grade 1chemistry Half Yearly Examination

8. Staying Engaged with Grade 1chemistry Half Yearly Examination
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 1chemistry Half Yearly Examination
9. Balancing eBooks and Physical Books Grade 1chemistry Half Yearly Examination
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 1chemistry Half Yearly Examination
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade 1chemistry Half Yearly Examination
 - Setting Reading Goals Grade 1chemistry Half Yearly Examination
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 1chemistry Half Yearly Examination
 - Fact-Checking eBook Content of Grade 1chemistry Half Yearly Examination
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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