

Test Unit 3 SCH3U #10

Name: _____

15 K 20 T 5 C 7 A Assessment of Learning (Product)

Multiple Choice: Circle the most correct answer. (10K)

1. What is the molar mass of AuCl_3 ?

- A) 96 g B) 130 g C) 232.5 g D) 303.6 g

2. Avogadro's number of representative particles is equal to one _____.

- A) kilogram B) gram C) Kelvin D) mole

3. How many moles of tungsten atoms are in 4.8×10^{22} atoms of tungsten?

- A) 8.0×10^2 moles B) 8.0×10^1 moles C) 1.3×10^{-1} moles D) 1.3×10^{-2} moles

4. Determine the mass in grams of 0.0489 mol cobalt.

- A) 0.00812 g B) 0.294 g C) 2.88 g D) 0.000830 g

5. What is the mass in grams of 1.02×10^{24} atoms manganese (Mn)?

- A) 0.112×10^1 g B) 0.169×10^6 g C) 9.30×10^{-1} g D) 9.30×10^6 g

6. Which of the following is/are true?

- i. Samples of water coming from different sources will still have the same percentage composition
- ii. The percentage composition of a compound can ONLY be calculated with knowledge of the compound's formula
- iii. Percent composition is the relative volume of each element in the compound
- iv. Two compounds can have identical percentage compositions

- A) i, ii B) i, ii, iii C) ii, iii D) i, iv E) iv only

7. Which of the following is a candidate for an empirical formula?

- A) $\text{C}_{24}\text{H}_{40}\text{O}_{22}$ B) N_2O_4 C) CH_3OH D) C_6H_6 E) $\text{Na}_2\text{S}_2\text{O}_6$

8. The mass percent of nitrogen, in nitric acid, HNO_3 , is

- A) 1.6% B) 11.1% C) 15.6% D) 22.2%

9. Hydrogen peroxide is known to contain 5.9% hydrogen and 94.1% oxygen, by mass. If its molar mass is 34.0 g/mol, its molecular formula is

- A) HO B) H_2O_2 C) HO_2 D) HNO_2

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1. Understanding the eBook Grade 11 Stoichiometry Test
 - The Rise of Digital Reading Grade 11 Stoichiometry Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 11 Stoichiometry Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 11 Stoichiometry Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 11 Stoichiometry Test
 - Personalized Recommendations
 - Grade 11 Stoichiometry Test User Reviews and Ratings
 - Grade 11 Stoichiometry Test and Bestseller Lists
5. Accessing Grade 11 Stoichiometry Test Free and Paid eBooks

- Grade 11 Stoichiometry Test Public Domain eBooks
- Grade 11 Stoichiometry Test eBook Subscription Services
- Grade 11 Stoichiometry Test Budget-Friendly Options
- 6. Navigating Grade 11 Stoichiometry Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 11 Stoichiometry Test Compatibility with Devices
 - Grade 11 Stoichiometry Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 11 Stoichiometry Test
 - Highlighting and Note-Taking Grade 11 Stoichiometry Test
 - Interactive Elements Grade 11 Stoichiometry Test
- 8. Staying Engaged with Grade 11 Stoichiometry Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 11 Stoichiometry Test
- 9. Balancing eBooks and Physical Books Grade 11 Stoichiometry Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 11 Stoichiometry Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 11 Stoichiometry Test
 - Setting Reading Goals Grade 11 Stoichiometry Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 11 Stoichiometry Test
 - Fact-Checking eBook Content of Grade 11 Stoichiometry Test
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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