

11 Chapter 11 Test, Form 1

Write the letter for the correct answer in the blank at the right of each question.

1. Find the area of parallelogram $ABCD$. Round to the nearest tenth.

A. 17.5 ft^2
B. 31.25 ft^2

C. 35 ft^2
D. 62.5 ft^2



1. _____

2. Find the area of parallelogram $WXYZ$. Round to the nearest tenth.

F. 27 in^2
G. 45 in^2

H. 63.6 in^2
J. 81 in^2



2. _____

3. What is the best classification of quadrilateral $PQRS$ with vertices $P(2, 2)$, $Q(-1, 2)$, $R(-1, -3)$, and $S(2, -3)$?

A. square
B. rectangle

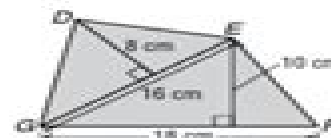
C. parallelogram
D. none of these

3. _____

4. Find the area of quadrilateral $DEFG$.

F. 154 cm^2
G. 218 cm^2

H. 244 cm^2
J. 308 cm^2



4. _____

5. Find the area of trapezoid $ABCD$ with vertices $A(1, -2)$, $B(5, -2)$, $C(4, 4)$, and $D(1, 4)$.

A. 6.5 units^2
B. 14 units^2

C. 21 units^2
D. 36 units^2

5. _____

6. Find the area of rhombus $ABCD$ with vertices $A(-1, 3)$, $B(3, 0)$, $C(-1, -3)$, and $D(-5, 0)$.

F. 8 units^2
G. 24 units^2

H. 26 units^2
J. 32 units^2

6. _____

7. Find the area of a regular octagon with a perimeter of 96 centimeters.

A. about 695.3 cm^2
B. about 576 cm^2

C. about 532 cm^2
D. about 119.3 cm^2

7. _____

8. Find the area of an equilateral triangle with a side length of 14 inches.

F. about 12.1 in^2
G. about 42 in^2

H. about 84.9 in^2
J. about 254.6 in^2

8. _____

9. Find the area of a circle with a circumference of 20π .

A. 400π
B. 314π

C. 200π
D. 100π

9. _____

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CL Gary



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web the titration of a weak acid with a strong base involves the direct transfer of protons from the weak acid to the hydroxide ion the reaction of the weak acid acetic acid with a strong base NaOH can be seen below in the reaction the

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web as seen in the chapter on the stoichiometry of chemical reactions titrations can be used to quantitatively analyze solutions for their acid or base concentrations in this section we will explore the underlying chemical equilibria that make acid base titrimetry a

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web in a titration a solution of known concentration the titrant is added to a solution of the substance being studied the analyte in an acid base titration the titrant is a strong base or a strong acid and the analyte is an acid or a base respectively

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