



gen chem test 1 review

definitions

- **atoms:** a submicroscopic particle that constitutes the fundamental building block of ordinary matter; the smallest identifiable unit of an element
- **molecules:** two or more atoms joined chemically in a specific geometric arrangement
- **hypothesis:** a tentative interpretation or explanation of an observation; a good hypothesis is falsifiable.
- **scientific law:** a brief statement or equation that summarizes past observations and predicts future ones
- **theory:** a proposed explanation for observations and laws, based on well-established and tested hypotheses, that presents a model for the way nature is and tries to show not merely what nature does but why
- **law of conservation of mass:** a law stating that matter is neither created nor destroyed in a chemical reaction
- **atomic theory:** the theory that each element is composed of tiny indestructible particles called atoms, that all atoms of a given element have the same mass and other properties, and that atoms combine in simple, whole number ratios to form compounds
- **solid:** a state of matter in which atoms or molecules are packed close to one another in fixed locations with definite volume
- **liquid:** a state of matter in which atoms or molecules pack about as close as they do in solid matter but are free to move relative to one another; liquids have a fixed volume but not a fixed shape

Gen Chem Sem Review Answers

David Baud



Gen Chem Sem Review Answers:

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