



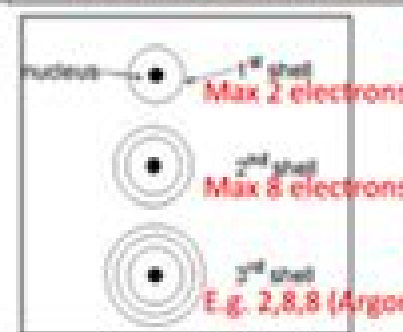
Periodic Table of the Elements

An atom:
Number of Protons = Number of Electrons
Mass number = Protons + Neutrons

C2 Chemistry

Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
E.g. Lithium 2,1 is a group 1 element.

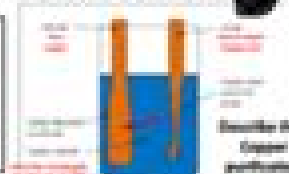


Protons and neutrons both have a mass of 1

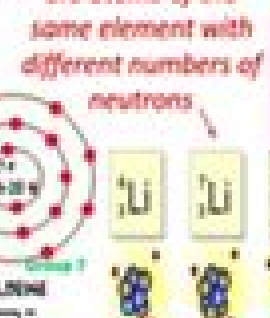
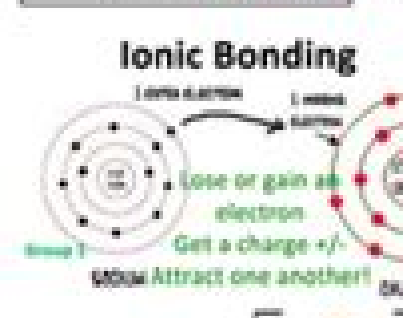
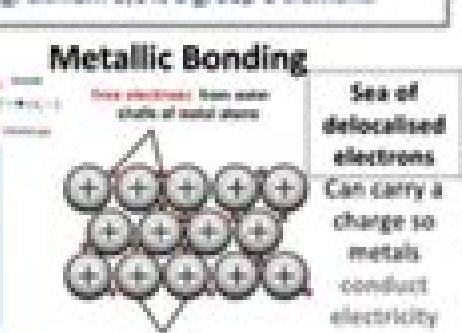
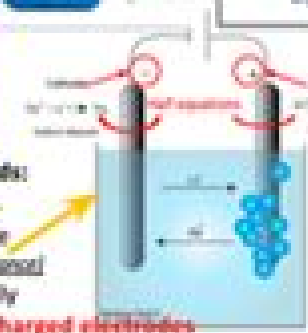
ISOTOPES
are atoms of the same element with different numbers of neutrons

OIL RIG

Oxidation is loss (of electrons)
Reduction is gain (of electrons)



Electrolysis:
Splitting up a substance using electricity



When there are 2 possible products
The less reactive forms at the electrode

Electrolysis of Brine

Used for: Disinfectants, Bleach and plastics

Used for: Margarine and HCl

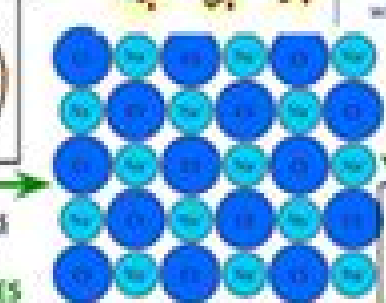
Electrolysis of Brine

Used for: Bleach, paper and soap

Metal ions only arranged in layers
ELECTROSTATIC ATTRACTION
between
+IONS and ELECTRONS

Metal ions slide over one another making them **MALLEABLE**

Metals conduct **HEAT** well



As a solid: High melting and Boiling Point
Conducts electricity when melted or dissolved:
Ions move freely and carry a charge

NANO-TECHNOLOGY
Means **REALLY REALLY REALLY** small!!!!

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 - The Rise of Digital Reading Gcse Additional Chemistry C2 Revision Booklet
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Additional Chemistry C2 Revision Booklet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Additional Chemistry C2 Revision Booklet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Additional Chemistry C2 Revision Booklet
 - Personalized Recommendations
 - Gcse Additional Chemistry C2 Revision Booklet User Reviews and Ratings
 - Gcse Additional Chemistry C2 Revision Booklet and Bestseller Lists
5. Accessing Gcse Additional Chemistry C2 Revision Booklet Free and Paid eBooks
 - Gcse Additional Chemistry C2 Revision Booklet Public Domain eBooks
 - Gcse Additional Chemistry C2 Revision Booklet eBook Subscription Services
 - Gcse Additional Chemistry C2 Revision Booklet Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
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 - Gcse Additional Chemistry C2 Revision Booklet Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Gcse Additional Chemistry C2 Revision Booklet
 - Highlighting and Note-Taking Gcse Additional Chemistry C2 Revision Booklet
 - Interactive Elements Gcse Additional Chemistry C2 Revision Booklet
8. Staying Engaged with Gcse Additional Chemistry C2 Revision Booklet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Additional Chemistry C2 Revision Booklet
9. Balancing eBooks and Physical Books Gcse Additional Chemistry C2 Revision Booklet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Additional Chemistry C2 Revision Booklet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gcse Additional Chemistry C2 Revision Booklet
 - Setting Reading Goals Gcse Additional Chemistry C2 Revision Booklet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gcse Additional Chemistry C2 Revision Booklet
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13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements

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