

2.5

ELEMENTS AND COMPOUNDS

Section Review

Objectives

1. Explain the difference between elements and a compound.
2. Distinguish between a mixture and a solution.
3. Identify the chemical symbols for elements and write a chemical formula for a compound.

Vocabulary

- | | |
|------------|----------------|
| 1. element | 2. compound |
| 3. mixture | 4. homogeneous |

Part A Completion

Fill in the blank with the word or phrase that best completes or describes each statement. Write your answer in the space provided.

- | | |
|---|-------------|
| 1. An element is made of _____ atoms. | 6. Element |
| 2. Compounds are made of _____ atoms. | 7. Mixture |
| 3. An element is made of _____ as a pure. | 8. Mixture |
| 4. Compounds are made of _____. | 9. Mixture |
| 5. Compounds are _____. | 10. Mixture |
| 11. Compounds are made of _____. | 11. Mixture |
| 12. Compounds are made of _____. | 12. Mixture |
| 13. Compounds are made of _____. | 13. Mixture |
| 14. Compounds are made of _____. | 14. Mixture |
| 15. Compounds are made of _____. | 15. Mixture |
| 16. Compounds are made of _____. | 16. Mixture |
| 17. Compounds are made of _____. | 17. Mixture |
| 18. Compounds are made of _____. | 18. Mixture |
| 19. Compounds are made of _____. | 19. Mixture |
| 20. Compounds are made of _____. | 20. Mixture |

Elements And Compounds Section Review

Roman Wölfel



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