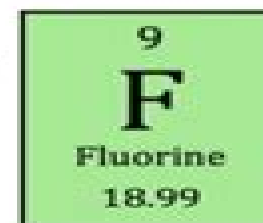
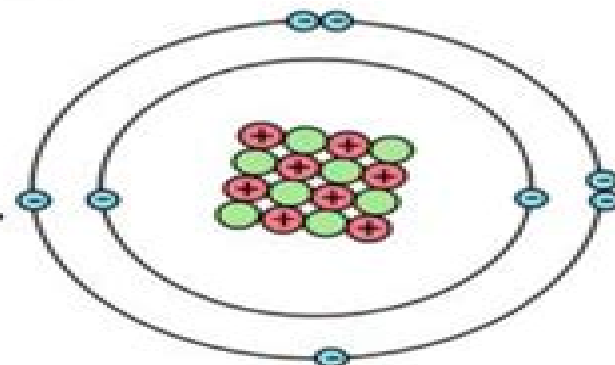


Atomic Theory – Unit Exam

For all of the following multiple-choice questions, choose only one correct answer:

- Which description below accurately describes the current model of the atom?
 - Electrons are distributed around the nucleus and have almost no mass compared to protons
 - Protons, electrons and neutrons are evenly distributed throughout all parts of the atom
 - The nucleus consists of protons, electrons and neutrons
 - The nucleus is made of electrons and protons
- Which of the following is **true** about subatomic particles?
 - Electrons are negatively charged and are the heaviest subatomic particle
 - The mass of a proton nearly equals the mass of a neutron
 - Protons are positively charged and the lightest subatomic particle
 - Neutrons are negatively charged and are the heaviest subatomic particle
- What would be the **mass** of the atom pictured to the right?
 - 7 amu
 - 8 amu
 - 16 amu
 - 24 amu
- How many **valence electrons** are shown in the atom on the right?
 - 2 valence electrons
 - 4 valence electrons
 - 6 valence electrons
 - 8 valence electrons
- Why are atoms electrically **neutral**?
 - The number of neutrons are equal to the number of protons and electrons.
 - The numbers of neutrons and electrons are equal and cancel each other out.
 - The numbers of protons and electrons are equal and cancel each other out.
 - The numbers of protons and neutrons are equal and cancel each other out.
- Based on the symbol to the right, what can be concluded about the element **fluorine**?
 - There are 19 atoms in a molecule of fluorine
 - There are 9 neutrons outside of the fluorine atom's nucleus
 - There are 9 protons in the nucleus of a fluorine atom
 - There are 9 valence electrons in the fluorine atom.



Electron Structure Test Review Guide

Daniel F McAuley



Electron Structure Test Review Guide:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **Electron Structure Test Review Guide**

In some sort of inundated with displays and the cacophony of instantaneous interaction, the profound power and emotional resonance of verbal art often fade in to obscurity, eclipsed by the continuous assault of noise and distractions. Yet, nestled within the musical pages of **Electron Structure Test Review Guide**, a charming perform of fictional splendor that pulses with raw feelings, lies an unique journey waiting to be embarked upon. Penned by a virtuoso wordsmith, this magical opus books viewers on a mental odyssey, softly exposing the latent potential and profound influence stuck within the delicate web of language. Within the heart-wrenching expanse of the evocative analysis, we shall embark upon an introspective exploration of the book is key themes, dissect their charming publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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