Name: _____ Date: _____

Electron configuration worksheet with answer key

Write the full electron configurations for each element.

Potassium (K)

1s 2s 2p 3s 3p 4s

· Sulfur (S)

1s²2s²2p⁶3s²3p⁴

Fluorine (F)

1s²2s²2p⁵

· Silicon (Si)

1s²2s²2p⁶3s²3p²

Write the shorthand (noble gas) electron configurations.

Calcium (Ca)

[Ar]4s2

Iron (Fe)

[Ar]4s²3d⁶

Nickel (Ni)

[Ar]4s²3d⁸

Copper (Cu)

[Ar]4s 3d 10

Identify elements based on their electron configurations.

- 1s² 2s² 2p⁶ 3s² 3p⁶ 4s¹ K (Potassium)
- [Ne] 3s² 3p³

P (Phosphorus)

[Ar] 4s² 3d⁸

(Nickel)

[Kr] 5s¹

(Rubidium)

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BM King

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