

4. Explain how the location of the electron bonds in Model 2 might account for the incredible hardness of a diamond.
5. Explain how the location of the "electron bonds" in Model 3 might account for the softness and flexibility of copper metal.
6. Both sugar ( $C_6H_{12}O_6$ ) and salt ( $NaCl$ ) dissolve in water but sugar does not conduct electricity. Use the models to explain how these two substances might be different once they dissolve in water.
7. Use the models to explain the following evidences
  - a. Molecular covalent substances are often gasses.
  - b. Ionic substances are brittle

# Electron Glue Answers

**SA Dillow**



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