# **Electron Configuration Practice Worksheet**

In the space below, write the unabbreviated electron configurations of the following elements:

1)	sodium
2)	iron
3)	bromine
4)	barium
5)	neptunium
	e space below, write the abbreviated electron configurations of the following ents:
6)	cobalt
7)	silver
8)	tellurium
9)	radium
10)	lawrencium
Dete	rmine what elements are denoted by the following electron configurations:
11)	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>4</sup>
12)	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup> 5s <sup>1</sup>
13)	[Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>3</sup>
14)	[Xe] 6s <sup>2</sup> 4f <sup>14</sup> 5d <sup>6</sup>
15)	[Rn] 7s <sup>2</sup> 5f <sup>11</sup>
Dete	rmine which of the following electron configurations are not valid:
16)	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 4d <sup>10</sup> 4p <sup>5</sup>
17)	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>3</sup> 3d <sup>5</sup>
18)	[Ra] 7s <sup>2</sup> 5f <sup>8</sup>
19)	[Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>5</sup>
20)	[Xe]

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**CL Gary** 

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