

floats when the pressure is 500 kPa. This stop is now cooled to 100 °C by rejecting heat to the surrounding at 20 °C. Find the total entropy generated in the process.

System = Water

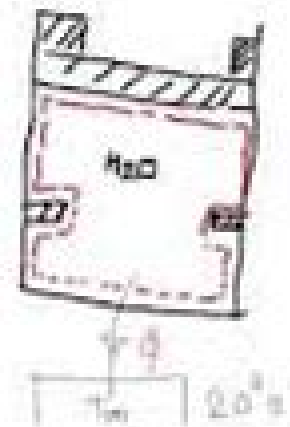
State 1  $\left\{ \begin{array}{l} P_1 = 1 \text{ MPa} \\ T_1 = 500^\circ\text{C} \end{array} \right\} m = \frac{V}{v}$

$\rho_{H_2O} = 958 \text{ kg/m}^3$   
 $V_1 = 3 \text{ m}^3$

final state,  $T = 100^\circ\text{C}$

When pressure inside the system is red. to 500 kPa

↳ floats  
 ↳ it starts to move.  
 $1 \text{ MPa} \rightarrow 500 \text{ kPa}$   
 $V = \text{const.}$



- When the system H.T.  $\uparrow$  to  $100^\circ\text{C}$   
 So, At this condition piston will hit the Lower

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This book comprises select proceedings of the International Conference on Innovations in Mechanical Engineering ICIME 2021 It presents innovative ideas and new findings in the field of mechanical engineering Various topics covered in this book are aerospace engineering automobile engineering thermal engineering renewable energy sources bio mechanics fluid mechanics MEMS mechatronics robotics CAD CAM CAE CFD design and optimization tribology materials engineering and metallurgy mimics surface engineering nanotechnology polymer science manufacturing production management industrial engineering and rapid prototyping This book will be useful for the students researchers and professionals working in the various areas of mechanical engineering

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