

Notes on Study-Questions

Changes in Kinetics:

Rate of Reaction:

Rate of Consumption: $-d[A]/dt$

Rate of Change: $d[A]/dt$

Rate of Accumulation: $\text{Accumulation Rate} = \text{Rate of Production} - \text{Rate of Consumption}$

Rate relationships when not:

- isothermal phase changes
- perfect piston/cylinder
- perfect fluid
- distinguish between solid and liquid regions
- isothermal heating/cooling
- ideal & nonideal gas behavior

Ex. 14.2.1

(a) Derive the following for the:

Reaction: $A \rightarrow B$ with A and B in a closed system at constant pressure and temperature.

Assumptions: (1) The system is closed and at constant pressure and temperature.

Derivation: (1) The system is closed and at constant pressure and temperature.

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Derivation: (1) The system is closed and at constant pressure and temperature.

(b) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

(c) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

(d) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

(e) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

(f) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

(g) What is the change in the total energy of the system?

Answer: (1) The system is closed and at constant pressure and temperature.

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S Ashworth



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