

Hydrocarbons

Methane	General formula for an alkane	General formula for an alkene	Ethene	Propene	Butene
Ethane	How to test between an alkene and alkane	Alkane or alkene? C ₂ H ₆ C ₃ H ₈ C ₄ H ₁₀ C ₅ H ₁₂	Show how ethene is polymerised		
Propane	Definition of a hydrocarbon		What is meant by saturated/unsaturated		
Butane	How ethanol is produced... ...from ethane ...by fermentation		Draw the monomer that makes this polymer $ \begin{array}{cccccc} \text{Br} & \text{H} & \text{Br} & \text{H} & \text{Br} & \text{H} \\ & & & & & \\ \text{H}-\text{C}- & \text{C}- & \text{C}- & \text{C}- & \text{C}- & \text{C}-\text{H} \\ & & & & & \\ \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} \end{array} $		
A way to remember the order of alkanes			Describe the process of cracking Process: Types of products produced: Conditions: Why it is done:		Complete the equations for cracking $\text{C}_{10}\text{H}_{22} \rightarrow \text{C}_2\text{H}_6 + \text{_____} + \text{C}_6\text{H}_{14}$ $\text{C}_8\text{H}_{18} \rightarrow \text{C}_3\text{H}_8 + \text{_____}$ The types of products produced:

Hydrocarbon Compounds Core Teaching Resources

Michael Hoelscher



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