



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

Grade 11 Term3 Physics Exemplar

Annelies Wilder-Smith



Grade 11 Term3 Physics Exemplar:

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Table of Contents Grade 11 Term3 Physics Exemplar

1. Understanding the eBook Grade 11 Term3 Physics Exemplar
 - The Rise of Digital Reading Grade 11 Term3 Physics Exemplar
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 11 Term3 Physics Exemplar
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 11 Term3 Physics Exemplar
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 11 Term3 Physics Exemplar
 - Personalized Recommendations
 - Grade 11 Term3 Physics Exemplar User Reviews and Ratings
 - Grade 11 Term3 Physics Exemplar and Bestseller Lists

5. Accessing Grade 11 Term3 Physics Exemplar Free and Paid eBooks
 - Grade 11 Term3 Physics Exemplar Public Domain eBooks
 - Grade 11 Term3 Physics Exemplar eBook Subscription Services
 - Grade 11 Term3 Physics Exemplar Budget-Friendly Options
6. Navigating Grade 11 Term3 Physics Exemplar eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 11 Term3 Physics Exemplar Compatibility with Devices
 - Grade 11 Term3 Physics Exemplar Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 11 Term3 Physics Exemplar
 - Highlighting and Note-Taking Grade 11 Term3 Physics Exemplar
 - Interactive Elements Grade 11 Term3 Physics Exemplar
8. Staying Engaged with Grade 11 Term3 Physics Exemplar
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 11 Term3 Physics Exemplar
9. Balancing eBooks and Physical Books Grade 11 Term3 Physics Exemplar
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 11 Term3 Physics Exemplar
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade 11 Term3 Physics Exemplar
 - Setting Reading Goals Grade 11 Term3 Physics Exemplar
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 11 Term3 Physics Exemplar
 - Fact-Checking eBook Content of Grade 11 Term3 Physics Exemplar
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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