

## Practice Test Ch 3

## Stoichiometry

Name Key No.     

- Remember, this is practice - Do NOT cheat yourself of finding out what you are capable of doing. Be sure you follow the testing conditions outlined below.
- DO NOT USE A CALCULATOR. AP Chem does not allow the use of a calculator for the MC part of the exam, so it is time to start practicing without one. You may use ONLY a periodic table.
- While you should practice working as fast as possible, it is more important at this point in the course, that you practice without a calculator, even if it slows you down. Look for the "easy math" - common factors and rough estimation - do not do "long division" to try to get exact values. Remember it is a MC test, use the answers.
- Mark which questions you would like to "go over" when we get to school in September.

1. Balance the following equation:



The balanced equation shows that 1.00 mole of  $\text{NH}_3$  requires      mole(s) of  $\text{O}_2$ :

- a. 0.57  
b. 1.25  
c. 1.33  
**d. 1.75**  
e. 3.5

$$1 \text{ mol NH}_3 \left( \frac{7 \text{ mol O}_2}{4 \text{ mol NH}_3} \right) =$$

2. Write a balanced equation for the combustion of acetaldehyde,
- $\text{CH}_3\text{CHO}$
- .



When properly balanced, the equation indicates that      mole(s) of  $\text{O}_2$  are required for each mole of  $\text{CH}_3\text{CHO}$ .

- a. 1  
b. 2  
**c. 2.5**  
d. 3  
e. 6

$$1 \text{ mol CH}_3\text{CHO} \left( \frac{5 \text{ mol O}_2}{2 \text{ mol CH}_3\text{CHO}} \right) =$$

3. What is the total mass of products formed when 16 grams of
- $\text{CH}_4$
- is burned with excess oxygen?

- a. 32 g  
b. 36 g  
c. 44 g  
d. 62 g  
**e. 80 g**



$$16 \text{ g CH}_4 \left( \frac{1 \text{ mol CH}_4}{16 \text{ g CH}_4} \right) \left( \frac{2 \text{ mol O}_2}{1 \text{ mol CH}_4} \right) \left( \frac{32 \text{ g O}_2}{1 \text{ mol O}_2} \right) =$$

4. Write a balanced equation for the combustion of propane,
- $\text{C}_3\text{H}_8$
- .



When balanced, the equation indicates that      moles of  $\text{O}_2$  are required for each mole of  $\text{C}_3\text{H}_8$ .

- a. 1.5  
b. 3  
c. 2.5  
**d. 5**  
e. 8

5. Balance the following equation with the SMALLEST WHOLE NUMBER COEFFICIENTS possible. Select the number that is the sum of the coefficients in the balanced equation:



- a. 3  
b. 5  
c. 6  
**d. 7**  
e. 8

6. Calculate the mass of hydrogen formed when 27 g of aluminum reacts with excess hydrochloric acid according to the balanced equation below:



- a. 1.5 g  
b. 2.0 g  
**c. 3.0 g**  
d. 6.0 g  
e. 12 g

$$27 \text{ g Al} \left( \frac{1 \text{ mol Al}}{27 \text{ g Al}} \right) \left( \frac{3 \text{ mol H}_2}{2 \text{ mol Al}} \right) \left( \frac{2 \text{ g H}_2}{1 \text{ mol H}_2} \right) =$$

7. How many grams of nitric acid,
- $\text{HNO}_3$
- , can be prepared from the reaction of 138 g of
- $\text{NO}_2$
- with 54.0 g
- $\text{H}_2\text{O}$
- according to the equation below?



- a. 92  
b. 108  
**c. 126**  
d. 189  
e. 279

$$138 \text{ g NO}_2 \left( \frac{1 \text{ mol NO}_2}{46 \text{ g NO}_2} \right) \left( \frac{2 \text{ mol HNO}_3}{3 \text{ mol NO}_2} \right) \left( \frac{63 \text{ g HNO}_3}{1 \text{ mol HNO}_3} \right) =$$

$$54 \text{ g H}_2\text{O} \left( \frac{1 \text{ mol H}_2\text{O}}{18 \text{ g H}_2\text{O}} \right) \left( \frac{2 \text{ mol HNO}_3}{1 \text{ mol H}_2\text{O}} \right) \left( \frac{63 \text{ g HNO}_3}{1 \text{ mol HNO}_3} \right) =$$

8. Which of the following statements is true?

- I. The molar mass of  $\text{CaCO}_3$  is 100.1 g  $\text{mol}^{-1}$ .  
II. 50 g of  $\text{CaCO}_3$  contains  $9 \times 10^{23}$  oxygen atoms.  
III. A 200 g sample of  $\text{CaCO}_3$  contains 2 moles of  $\text{CaCO}_3$ .

- a. I only  
b. II only  
c. III only  
**d. I and III only**  
e. I, II, and III

$$\frac{40}{100} = 0.4$$

# General Chemistry Stoichiometry Test Answers

**DP Hallahan**



## **General Chemistry Stoichiometry Test Answers:**

General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

MCAT General Chemistry Review 2023-2024 Kaplan Test Prep, 2022-08-02 Kaplan's MCAT General Chemistry Review 2023 2024 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined Prepping for the MCAT is a true challenge Kaplan can be your partner along the way offering guidance on where to focus your efforts and how to organize your review This book has been updated to match the AAMC's guidelines precisely no more worrying about whether your MCAT review is comprehensive The Most Practice More than 350 questions in the book and access to even more online more practice than any other MCAT general chemistry book on the market The Best Practice Comprehensive general chemistry subject review is written by top rated award winning Kaplan instructors Full color 3 D illustrations from Scientific American charts graphs and diagrams help turn even the most complex science into easy to visualize concepts All material is vetted by editors with advanced science degrees and by a medical doctor Online resources including a full length practice test help you practice in the same computer based format you'll see on Test Day Expert Guidance High yield badges throughout the book identify the topics most frequently tested by the AAMC We know the test The Kaplan MCAT team has spent years studying every MCAT related document available Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test

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College Board has discontinued the SAT Subject Tests in the US The tests will be available outside the US in June 2021 and then be discontinued Barron's SAT Subject Test Chemistry with 7 Practice Tests features in depth review of all topics on the exam and full length practice tests in the book and online This edition includes One full length diagnostic test to help you assess your strengths and weaknesses Comprehensive review of all topics on the exam including introductory chemistry atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory Four full length practice tests that reflect the actual SAT Subject Test Chemistry exam in length question types and degree of difficulty Two full length online practice tests with answer explanations and automated scoring Appendices which include the periodic table important equation constant and data tables and a glossary of chemistry terms *AP Chemistry Premium, 2024: 6 Practice Tests + Comprehensive Review + Online Practice* Neil D. Jespersen, Pamela Kerrigan, 2023-07-04 Always study with the most up to date prep Look for AP Chemistry Premium 2025 Prep Book with 6 Practice Tests Comprehensive Review Online Practice ISBN 9781506291802 on sale July 2 2024 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product DAT Aftab S. Hassan, 1993 *Master the Natural Sciences CLEP Test* Peterson's, 2012-04-30 Natural Sciences part of Peterson's Master the CLEP offers a review of the subject matter you need to know to master the scientific concepts that are tested on the CLEP Natural Sciences examination You will learn about evolution and classification cellular and molecular biology organisms and heredity ecology and population biology as well as the atom elements and reactions thermodynamics electromagnetism the structure of the universe and Earth's history and systems To help you pinpoint in which areas you may require further practice this review offers a 50 question pre test overview practice questions and a 50 question post test You will find in depth answer explanations for every question presented in this guide **Master the College Mathematics CLEP Test**

Peterson's, 2012-04-30 College Mathematics part of Peterson's Master the CLEP is meant to fully prepare you for the CLEP College Mathematics general examination It offers an extensive overview of the following subject areas real number systems sets probability and statistics logic functions and their graphs and additional topics from algebra and geometry You will find numerous practice question sets throughout the subject review In addition there are a pre and a post test with 50 questions each that will help you in your preparation for this exam All practice questions come with detailed answer explanations

*Chemistry Education* Javier García-Martínez, Elena Serrano-Torregrosa, 2015-02-17 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable

future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students

**Computer-assisted Instruction in Chemistry: General approach** Harry B. Mark, Hubert C. MacDonald, 1974      **MCAT 528 Kaplan Test Prep**, 2016-07-05

More people get into medical school with a Kaplan MCAT course than all major courses combined Now the same results are available with Kaplan s MCAT 528 This book features thorough subject review more questions than any competitor and the highest yield questions available The commentary and instruction come directly from Kaplan MCAT experts and include targeted focus on the most tested concepts MCAT 528 offers UNPARALLELED MCAT KNOWLEDGE The Kaplan MCAT team has spent years studying every MCAT related document available In conjunction with our expert psychometricians the Kaplan team is able to ensure the accuracy and realism of our practice materials THOROUGH SUBJECT REVIEW Written by top rated award winning Kaplan instructors all material has been vetted by editors with advanced science degrees and by a medical doctor EXPANDED CONTENT THROUGHOUT As the MCAT has continued to develop this book has been updated continuously to match the AAMC s guidelines precisely no more worrying if your prep is comprehensive STAR RATINGS FOR EVERY SUBJECT New for the 3rd Edition of MCAT 528 every topic is assigned a star rating informed by Kaplan s decades of MCAT experience and facts straight from the testmaker of how important it will be to your score on the real exam MORE PRACTICE THAN THE COMPETITION With 500 questions throughout the book and access to a full length practice test online MCAT 528 has more practice than any other advanced MCAT book on the market ONLINE COMPANION One practice test and additional online resources help augment content studying The MCAT is a computer based test so practicing in the same format as Test Day is key KAPLAN S MCAT REPUTATION Kaplan is a leader in the MCAT prep market and twice as many doctors prepared for the MCAT with Kaplan than with any other course UTILITY MCAT 528 can be used alone or with the other companion books in Kaplan s MCAT Review series Doctors refers to US MDs who were licensed between 2001 2010 and used a fee based course to prepare for the MCAT The AlphaDetail Inc online study for Kaplan was conducted between Nov 10 Dec 9 2010 among 763 US licensed MDs of whom 462 took the MCAT and used a fee based course to prepare for it



## Unveiling the Energy of Verbal Art: An Emotional Sojourn through **General Chemistry Stoichiometry Test Answers**

In a global inundated with screens and the cacophony of quick conversation, the profound power and psychological resonance of verbal art usually disappear into obscurity, eclipsed by the regular barrage of sound and distractions. Yet, situated within the lyrical pages of **General Chemistry Stoichiometry Test Answers**, a captivating function of literary beauty that impulses with natural emotions, lies an unforgettable journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, that mesmerizing opus courses readers on a mental odyssey, softly revealing the latent possible and profound affect stuck within the complex internet of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book's main themes, dissect its charming writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers' souls.

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