Section III: Problems (Answers to odds in back of the text book)

Practice Problems:

Chem Practice Problems:

1. Ammonia is a strong household cleaner. What is the atomic mass of one molecule?

17.04amu

- 2. The double replacement ...
- 3. A bottle of lead II sulfate contains 158.1g of the compound. How many moles of the compound are present?

0.5213 mol PbSO₄

- 4. Find the
- Determine the number of formula units that are in 1.48 mol NaF 8.91 x 10²³ units NaOH
- How many formula ...
- How many atoms are in 1.25 mol of oxygen gas?
 1.27 x 10²⁴ atoms O
- If you burned 6.10 x 10²⁴ ...
- A container with a volume of 101L contains how many moles of argon gas at STP?
 4.51 mol Ar
- 10. Find the percent composition ...
- 11. A sample of an unknown compound with a mass of 0.847g has the following composition 50.51% F & 49.49% Fe (m/m). When the compound is decomposed into its elements, what is the contributing mass of each? Write the decomposition equation after reassembling the empirical formula.
 .428g F & .419g Fe

ALOG F & AIST FE

.0225mol F = 3 & .00751 mol Fe = 1; 2FeF₃ → 2Fe + 3F₂ .00751 mol .00751 mol

- 12. Find the molecular formula ...
- 13. A 50.51g sample of made of phosphorus and chlorine is decomposed. Analysis of the products showed that 11.39g of phosphorus atoms were produced.
 - a. What is the empirical formula of the compound? 11.39/30.97 = <u>.368mol</u> P & (50.51g - 11.39g)/35.45g = .551mol Cl .368 mol .368 mol

General Chemistry Homework Guide

United States. Division of Vocational Education

General Chemistry Homework Guide:

General Chemistry Homework Guide Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **General Chemistry Homework Guide**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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