

gen chem test 1 review

definitions

- atoms: a submicroscopic particle that constitutes the fundamental building block of ordinary matter; the smallest identifiable unit of an element
- molecules: two or more atoms joined chemically in a specific geometric arrangement
- hypothesis: a tentative interpretation or explanation of an observation; a good hypothesis is falsifiable.
- scientific law: a brief statement or equation that summarizes past observations and predicts future ones
- theory: a proposed explanation for observations and laws, based on wellestablished and tested hypotheses, that presents a model for the way nature is and tries to show not merely what nature dues but why
- law of conservation of mass: a law stating that matter is neither created nor destroyed in a chemical reaction
- atomic theory: the theory that each element is composed of tiny indestructible
 particles called atoms, that all atoms of a given element have the same mass and
 other properties, and that atoms combine in simple, whole number ratios to form
 compounds
- solid: a state of matter in which atoms or molecules are packed close to one another in fixed locations with definite volume
- liquid: a state of matter in which atoms or molecules pack about as close as they
 do in solid matter but are free to move relative to one another; liquids have a fixed
 volume but not a fixed shape

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J Spring

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